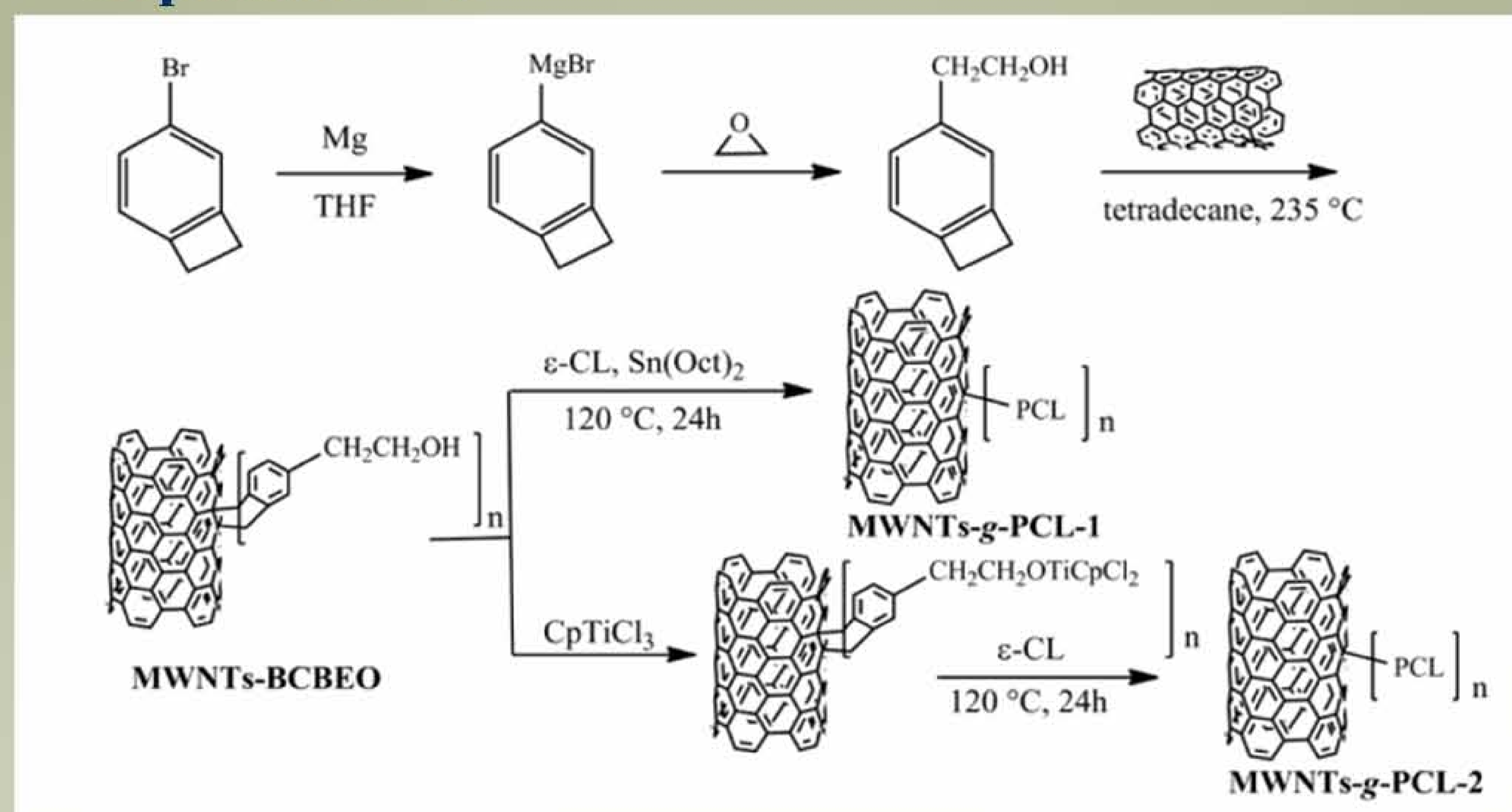


Introduction: Carbon nanotubes (CNTs) have many extraordinary electronic, mechanical, thermal and chemical properties, while for the strong van de Waals interaction CNTs exhibit a tendency of aggregation. Covalently attaching poly (ϵ -caprolactone) (PCL) onto the surface of CNTs could greatly improve the dispersity of CNTs in some organic solvents. Because of the one-end fixation of PCL chains, the crystallization (both bulk and solution) behaviors have their unique characteristics.

Experiment^[1]:



Scheme 1 Covalent functionalization of MWNTs using a [4 + 2] Diels-Alder cycloaddition reaction.

Results:

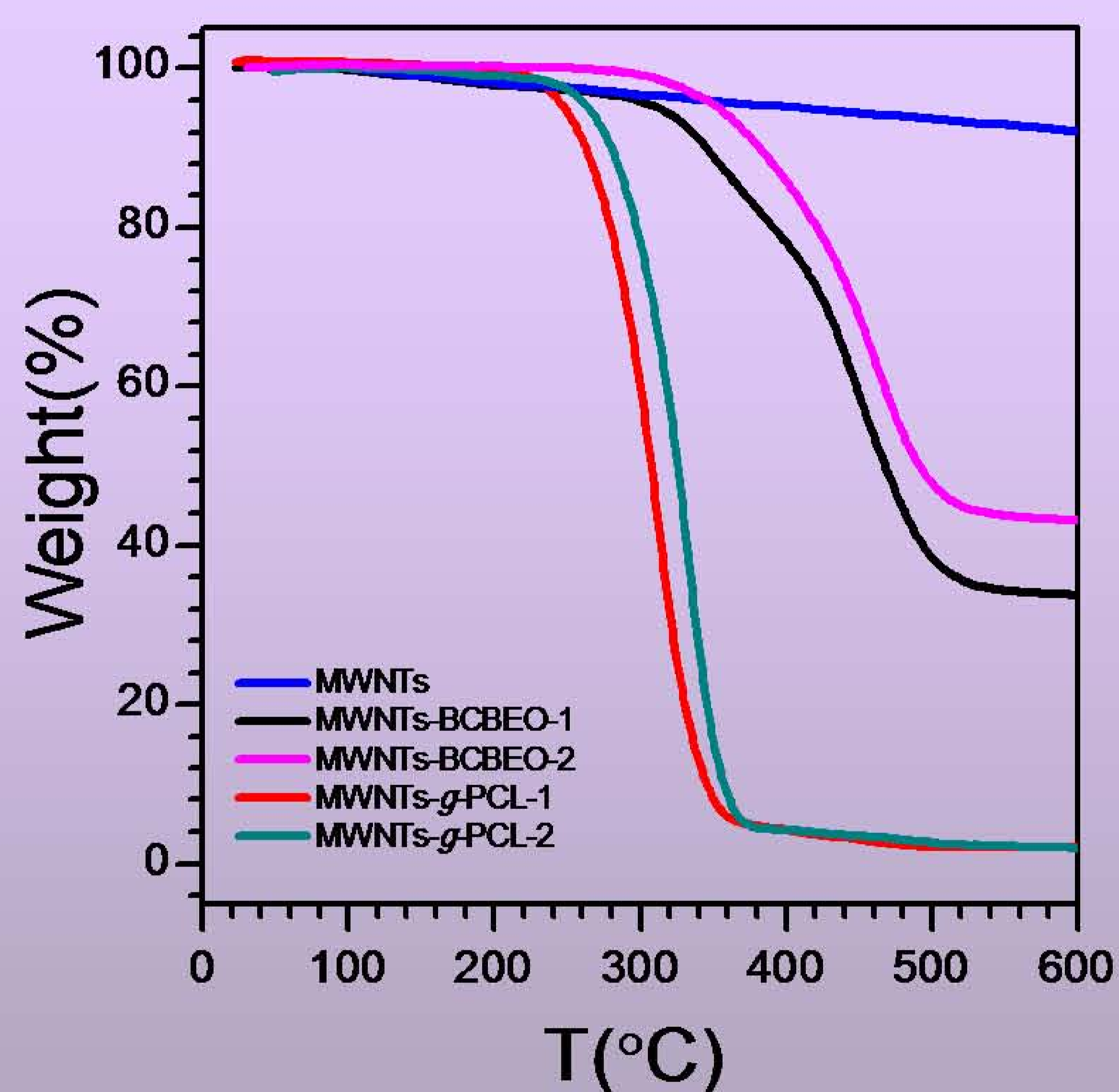


Figure 1 TGA curves of pristine MWNTs, MWNTs-BCBEOs and MWNTs-g-PCLs^[2].

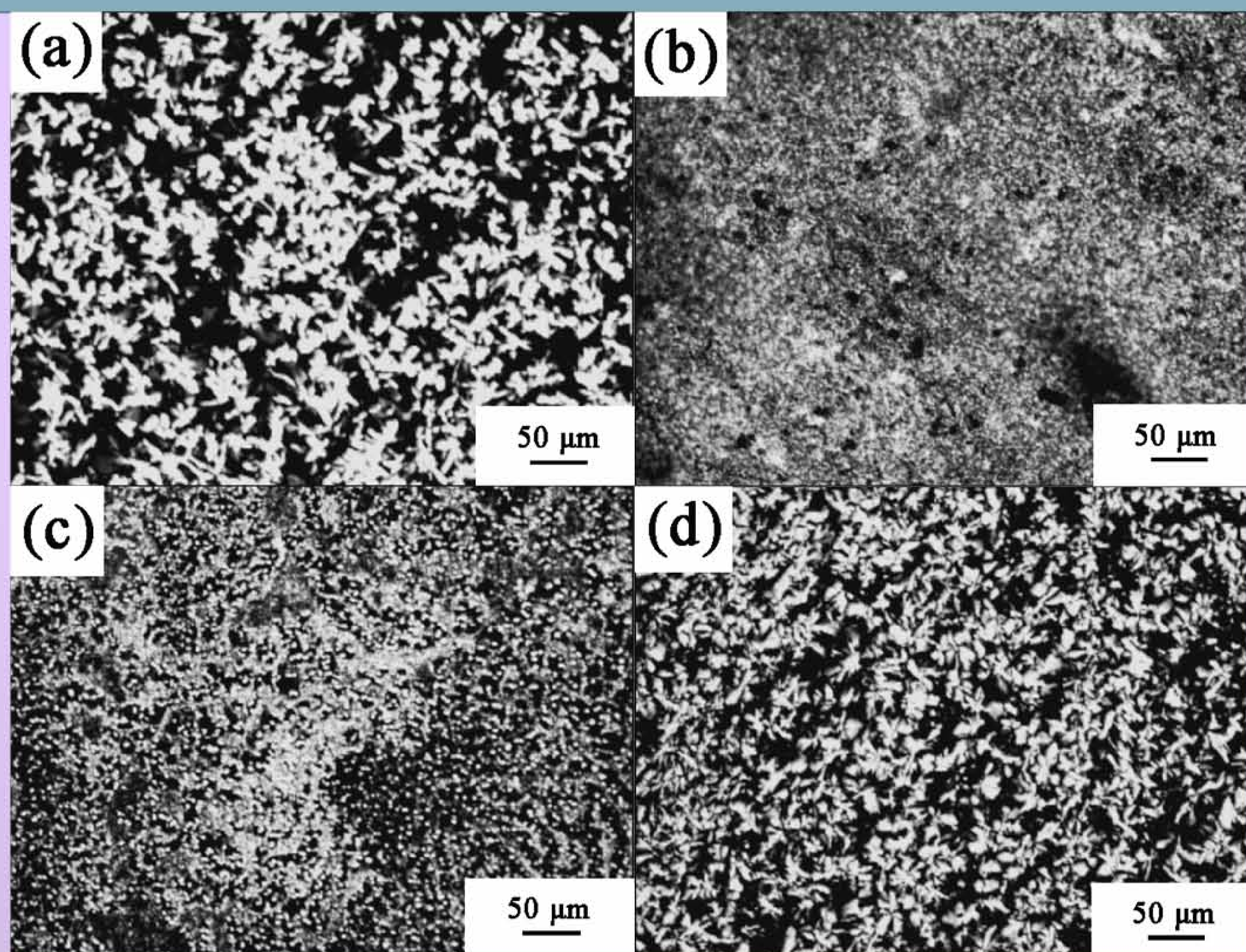


Figure 2 POM images of different samples after melting at 100 °C and then cooling to room temperature at a cooling rate of 2 °C/min. (a) neat PCL; (b) MWNTs-g-PCL-1; (c) MWNTs-g-PCL-2; (d) PCL/MWNTs blend.

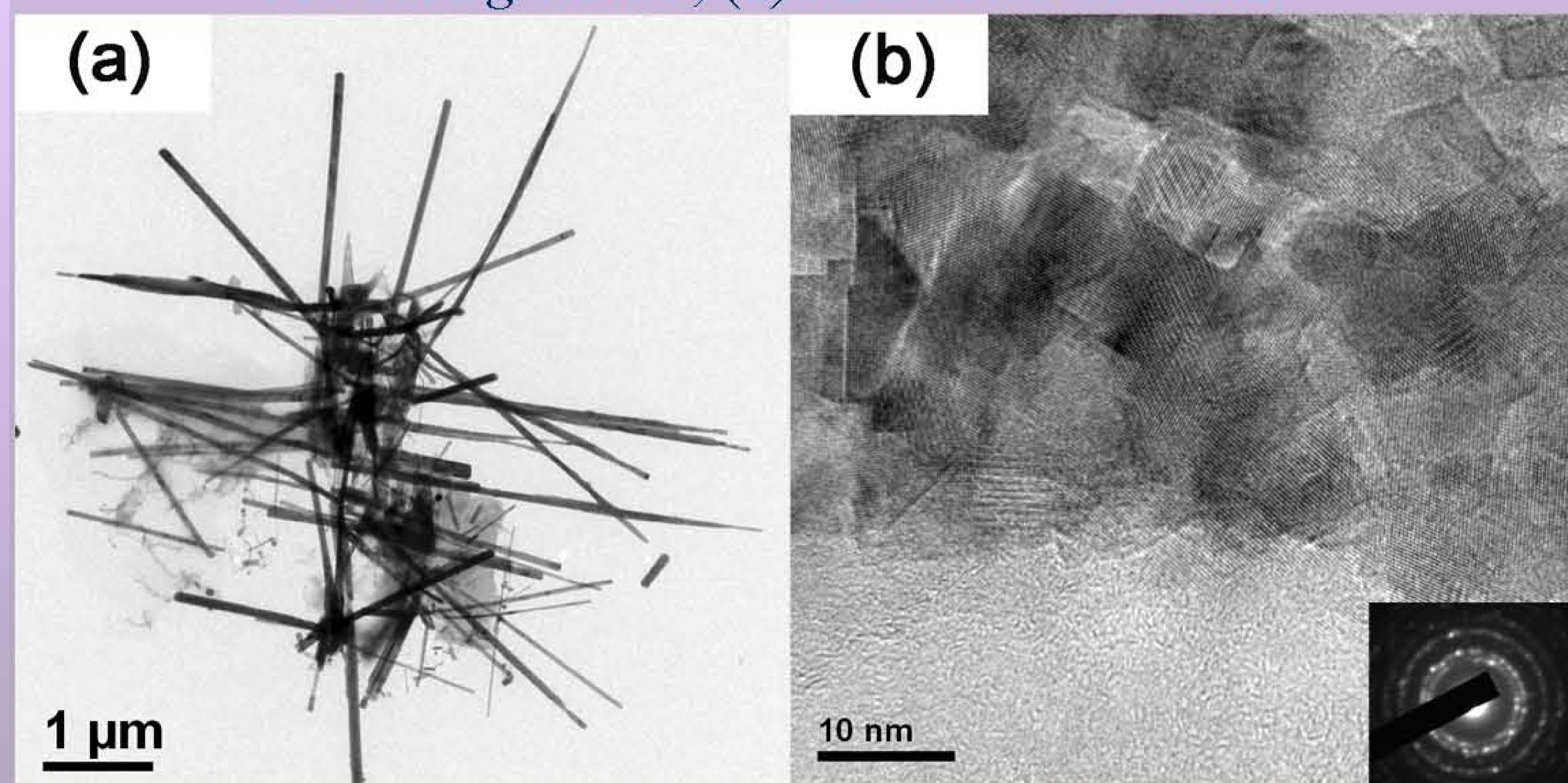


Figure 3 (a) TEM images of solution crystallized MWNTs-g-PCL, (b) HRTEM image of solution crystallized MWNTs-g-PCL and SAED of the same area.

Conclusion:

1. A series of MWNTs-g-PCLs with high PCL grafting density were successfully synthesized.
2. Comparing with neat PCL and PCL/MWNTs blend, MWNTs-g-PCLs have much stronger heterogeneous nucleation, as well as some confinement effect during bulk crystallization.
3. In *n*-hexanol solution, PCL chains in MWNTs-g-PCLs could crystallize on the surface of MWNTs.

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