



An Indandione Functionalized Tetraphenylethylene Fluorogen for the Detection of Arginine and Lysine

Jiaqi Tong(11129019),^a Anjun Qin,^a Jing Zhi Sun^{*a} and Ben Zhong Tang^{*a,b}

^a MOE Key Laboratory of Macromolecular Synthesis and Functionalization, Department of Polymer Science & Engineering, Zhejiang University, Hangzhou, China

^b Department of Chemistry, Institute for Advanced Study, Institute of Molecular Functional Materials, The Hong Kong University of Science & Technology, Clear Water Bay, Kowloon, Hong Kong, China

INTRODUCTION

As two of the most alkaline common amino acids, both arginine (Arg) and lysine (Lys) play unique roles in a series of biological processes, and the sensing of these two special amino acids is of great fundamental and practical significance.^{1,2} A few techniques, such as high-performance liquid chromatography and electrochemical analysis, have been concreted in practice and reported in literatures. Here, we present a concept-proof work of using a fluorescence probe to detection Arg and Lys over other amino acids based on an aggregation-induced emission (AIE) fluorogen.³

RESULTS AND DISCUSSION

1. Selectivity

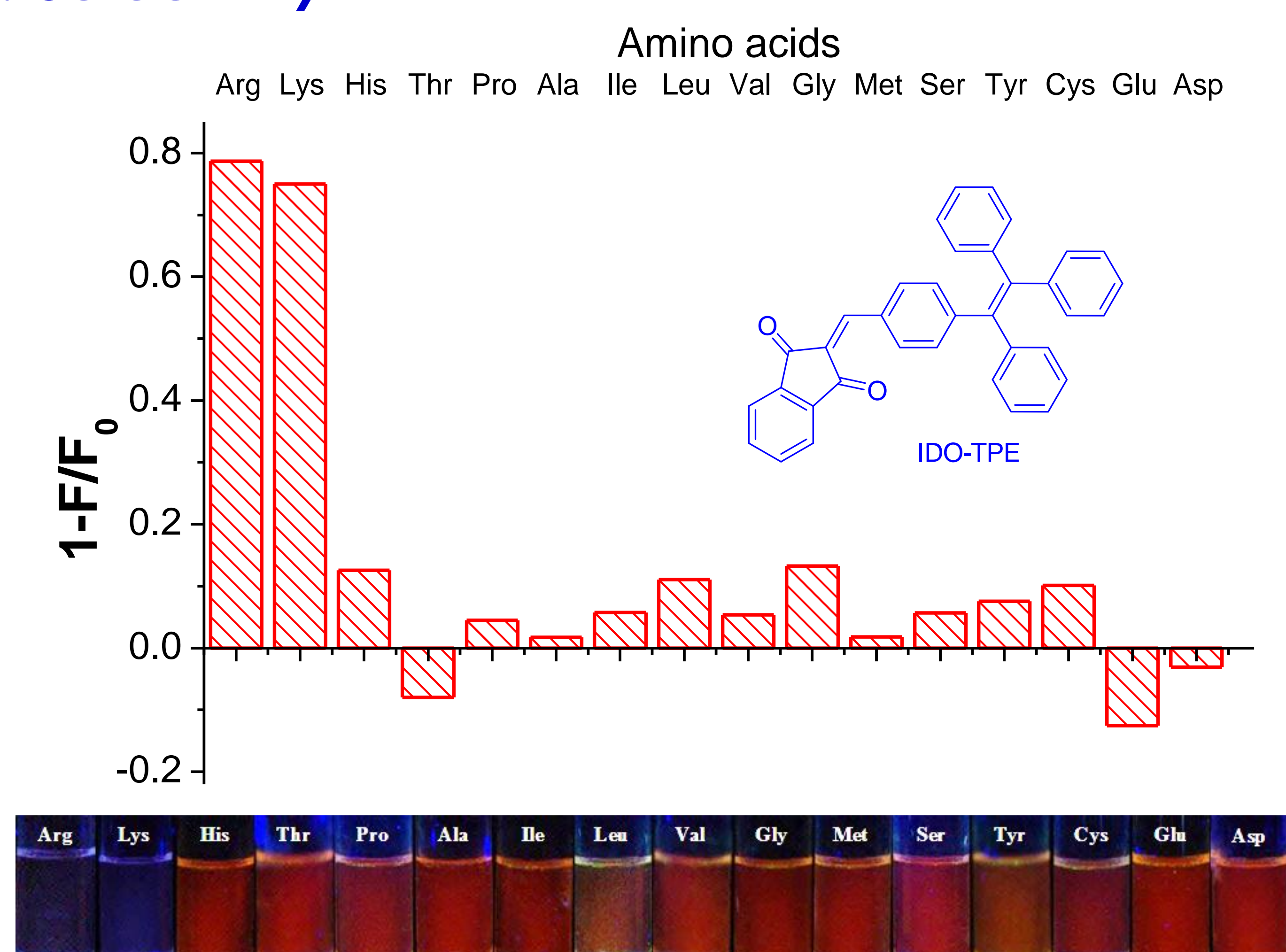


Figure 1. Specific response of IDO-TPE to diverse amino acids.

2. Time and Concentration Dependence

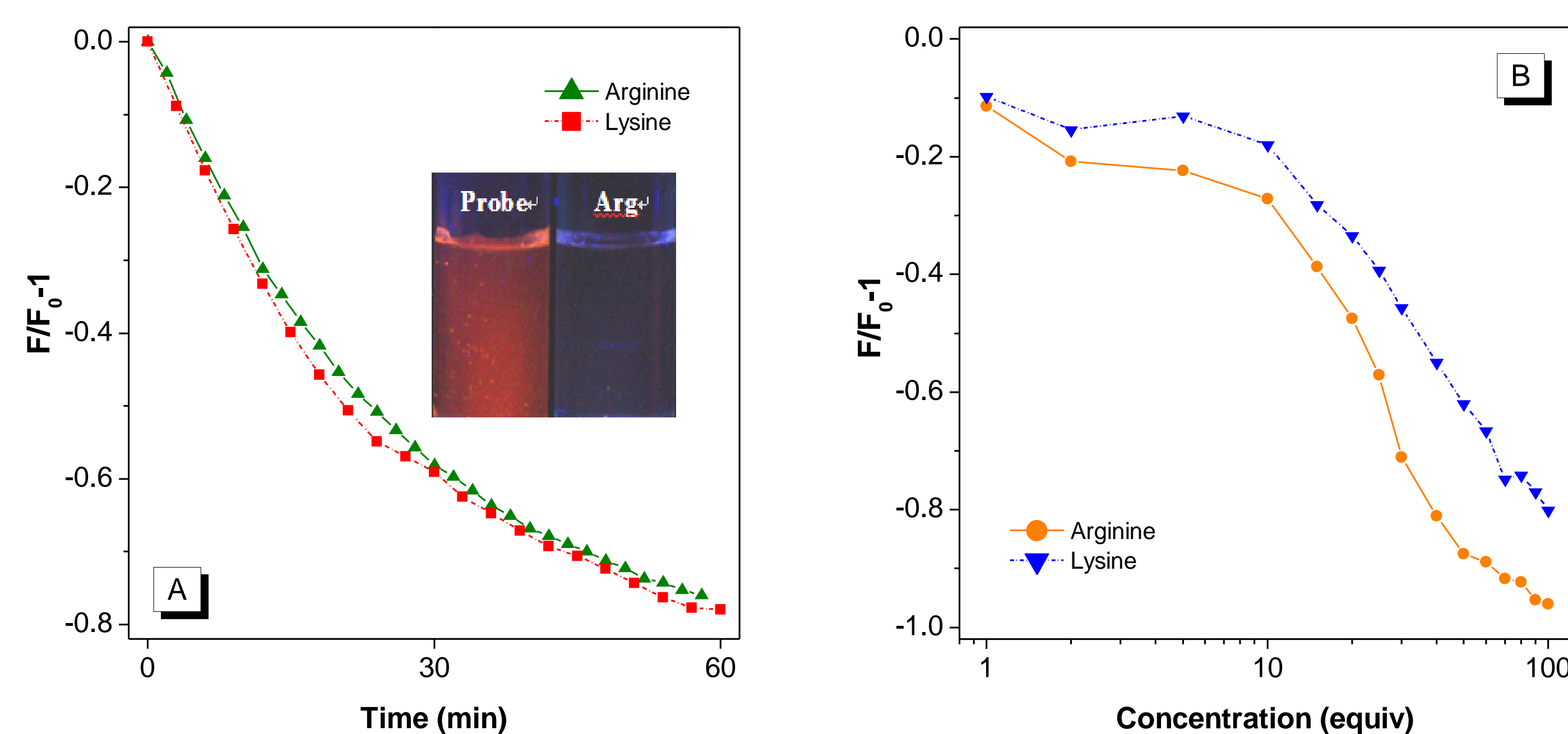
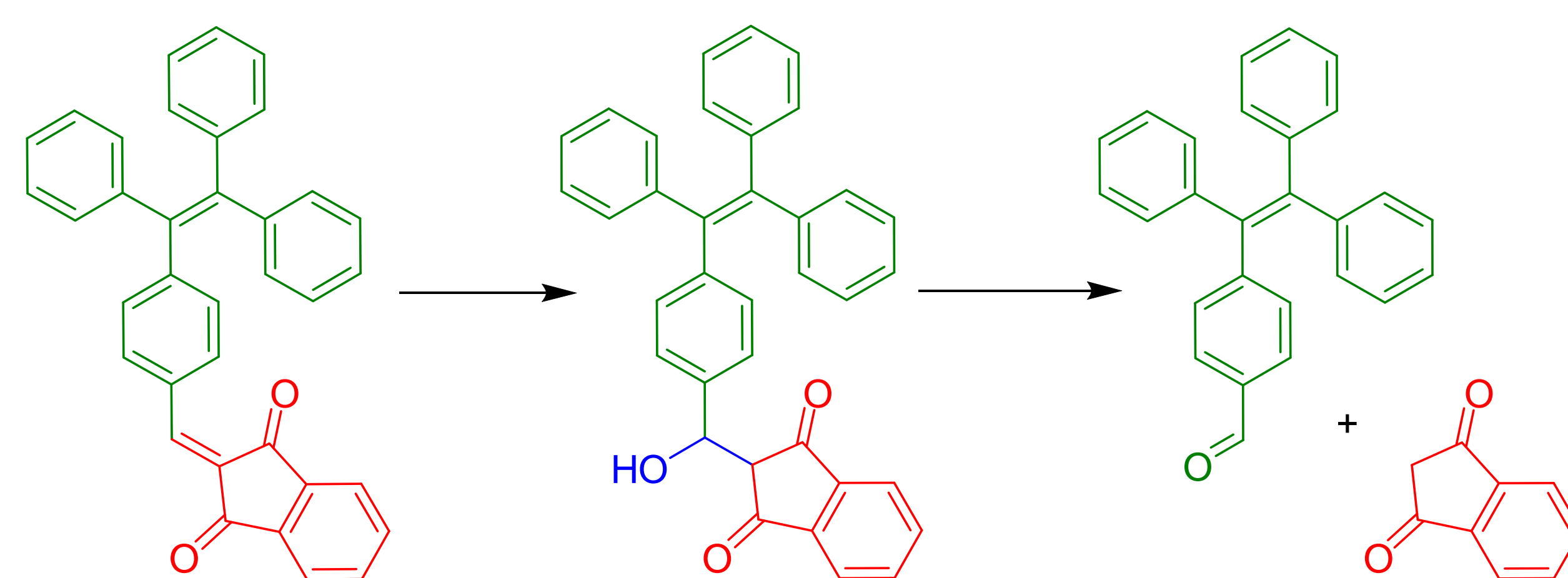


Figure 2. (A) Time dependent FL intensity of IDO-TPE + Arg and IDO-TPE + Lys; (B) FL intensity of IDO-TPE with different concentrations of Arg and Lys. Concentration of IDO-TPE: 10 μ M; Excitation wavelength: 420 nm; in THF/water mixture.

3. Proposed Mechanism



Scheme 1. Proposed mechanism of the detection of Arg and Lys by IDO-TPE.

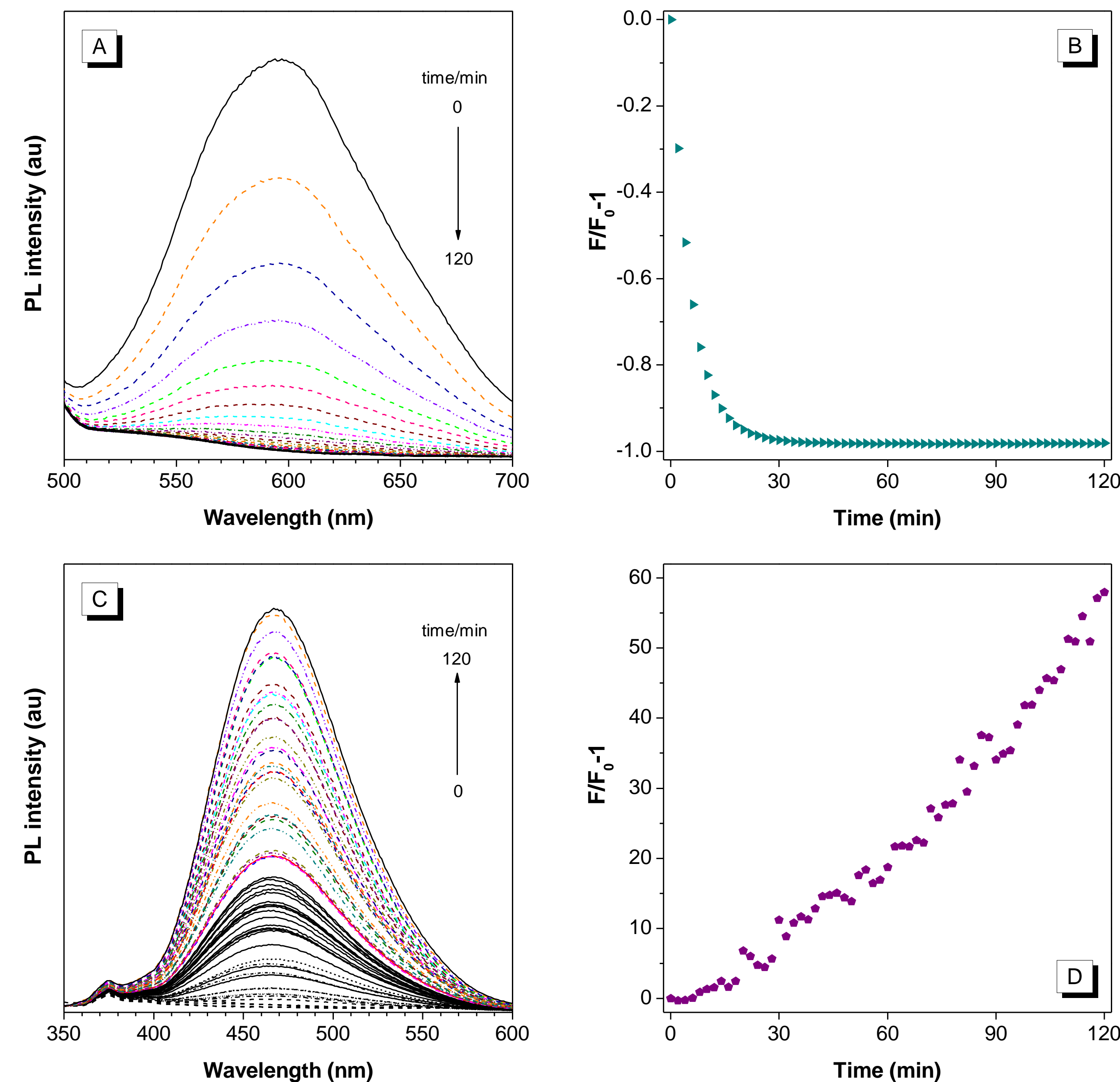


Figure 3. Time dependent FL spectra of IDO-TPE with an excitation wavelength at (A) 420 nm and (C) 330 nm in THF/buffer (pH 9.0); (B) and (D) corresponding plots of the changes in FL intensity for (A) and (C).

SUMMARY

A fluorescent probe of IDO-TPE has been used to detect Arg and Lys over other amino acids. With the addition of Arg or Lys into buffer solution of IDO-TPE, the orange emission band declines and a green band, assigned to the emission from TPE moiety, grows up. The change in emission color and intensity is explained by the reaction of IDO-TPE with OH^- , which destroys the conjugation between the indandione and the TPE moiety and recovers the fluorescence of the TPE moiety.

ACKNOWLEDGMENTS

This work was partially supported by the key project of the Ministry of Science and Technology of China (2013CB874304).

References

- [1] D. Juan *et al.* *Curr. Pharm. Biotechnol.*, **2009**, 10, 609.
- [2] P. C. Bevilacqua *et al.* *Acc. Chem. Res.*, **2011**, 44, 1270.
- [3] Y. Hong, and B. Z. Tang *et al.* *Chem. Soc. Rev.*, **2011**, 40, 5361.