

# Chemical Safety (II)

## Nanosafety, Waste Management & Emergency Response

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(HSEO)

HKUST

# Important Reminder

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Safety training courses provided by HSEO only cover BASIC safety principles. Your supervisor is responsible for your job-specific safety training, including operational safety procedures, departmental safety policies, etc.

# Presentation Outlines

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- Nanomaterials and nanosafety
- Chemical waste legal requirements
- HKUST chemical waste procedures
- Incidents and lessons learned
- Waste minimization
- Hazardous Material Inventory
- Emergency response
- Information Source

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# Nanomaterials & Nanosafety

# Different Nano-Related Materials

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- Materials with engineered nanostructure, e.g. zeolite with designed pore features
- Silicon-based nanomaterial or nanostructure, e.g. nanoelectronics
- Nanomaterials with one dimension not in nanoscale (i.e.  $>100$  nm), e.g. nanotube
- Nanoparticles (diameter  $<100$  nm)

# Origins of Nanoparticles

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- Naturally occurring, e.g. volcanic activities, wild fires
- Anthropogenic incidental nanoparticles, e.g. engines, incinerators
  - Also known as ultrafine particles,  $<100$  nm in AED
- Engineered nanoparticles
  - At least one dimension  $<100$  nm
  - Relatively uniform in size, monodispersed

# Unique Properties of Engineered Nanoparticles

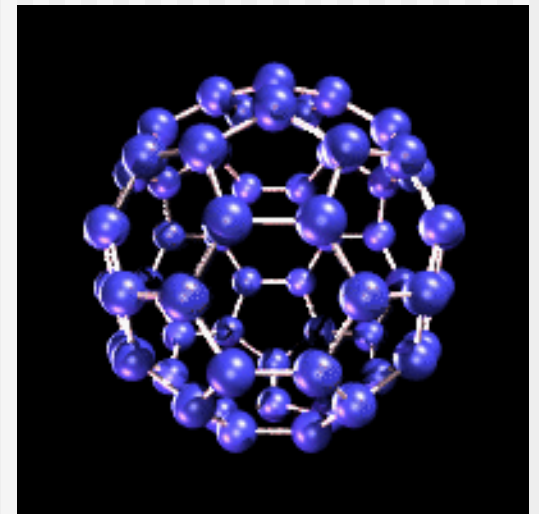
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- Novel physicochemical properties not found in parent materials
- Much higher reactivity, both chemical and biological
- Precisely designed molecules for specific purposes

# Buckyballs ( $C_{60}$ )

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- Extremely stable
- Can trap other atom or molecule inside
- Possible application
  - Superconducting material
  - Drug delivery system
  - Lubricant
  - Catalyst

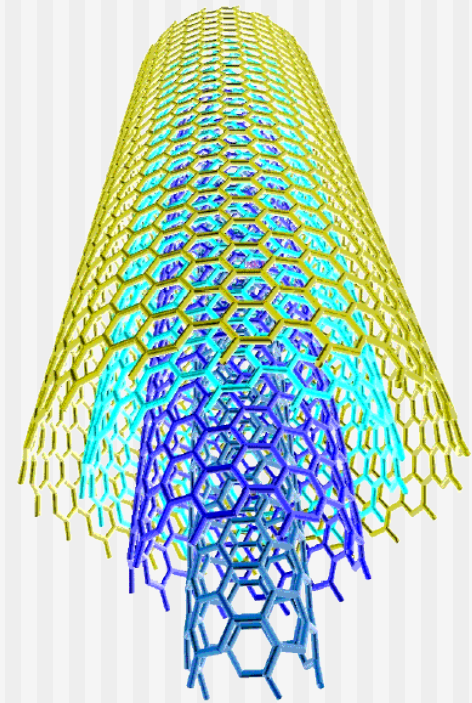




# Carbon Nanotubes

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- Field emitters in flat plane display
- 100X strength, 1/6 weight of steel, mechanically stronger materials
- Ultra-miniaturized electronics, one candidate of nonsilicon microchips



# Metal and Metal Oxide Nanoparticles

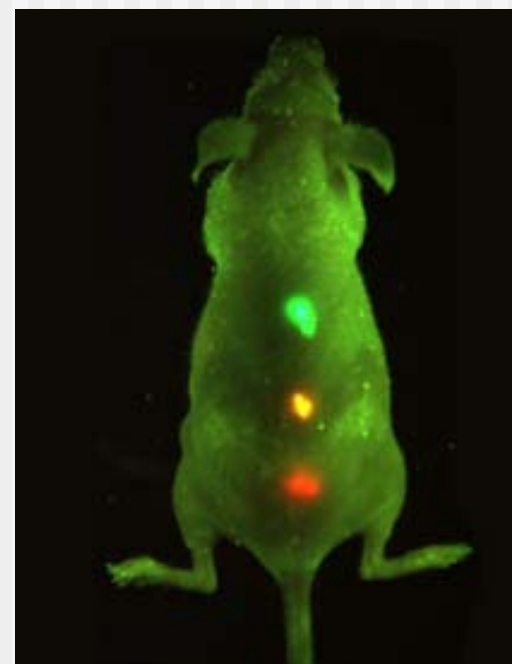
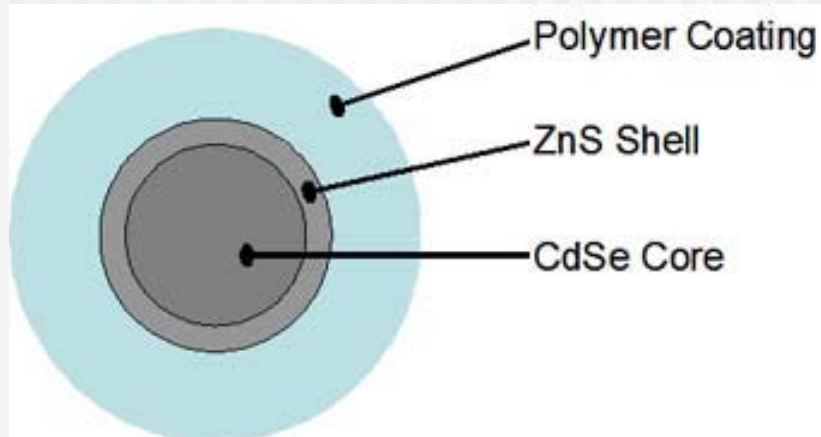
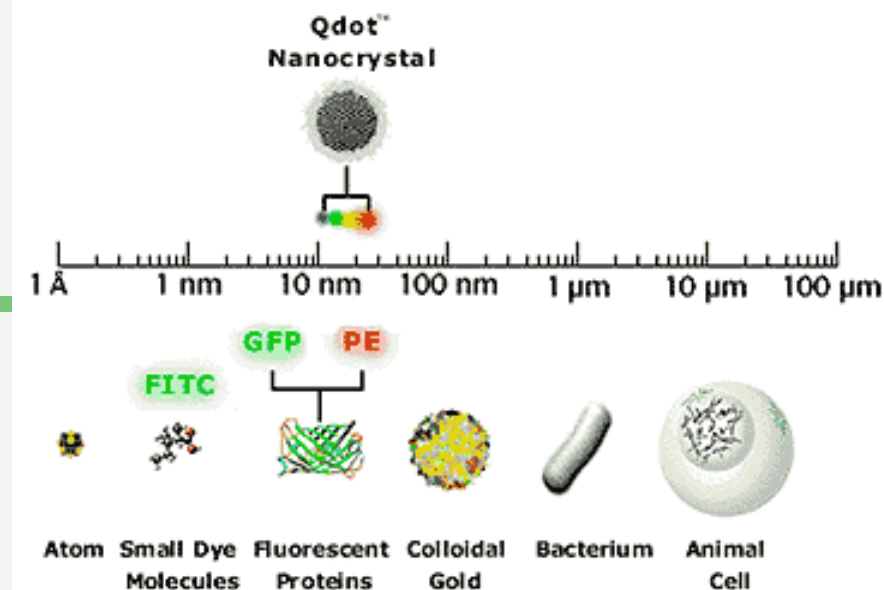
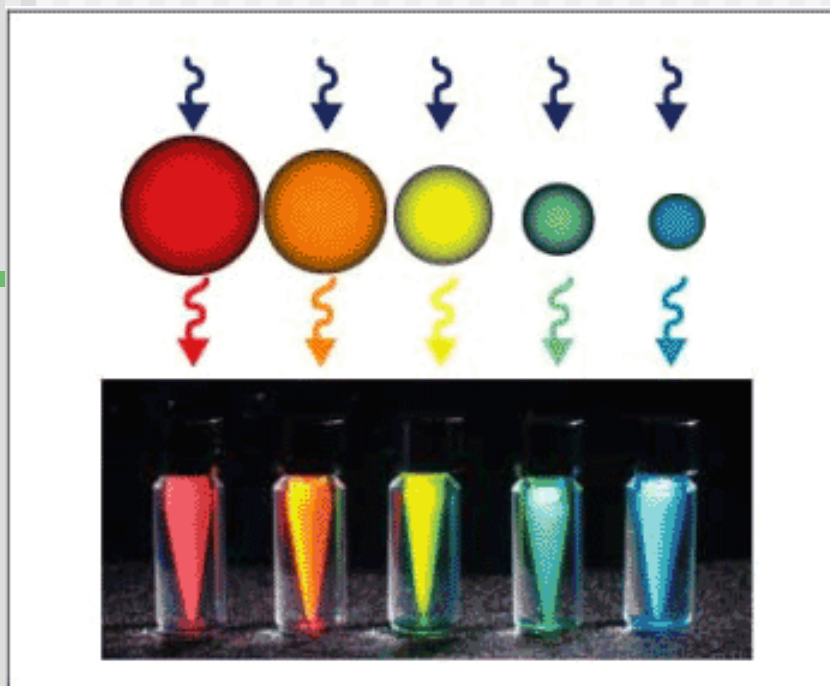
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- $\text{TiO}_2$  for photochemical oxidation, self-cleaning paints, etc
- $\text{ZnO}$  in cosmetics, sunblock
- Fe for oxidation of contaminants in groundwater
- $\text{CuO}$  for antimicrobial agent
- $\text{Fe}_2\text{O}_3$  as contrast agent for MRI

# Quantum Dots (Q-Dots)

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- Nanoparticles of semiconducting materials
- Size so small it interfere with quantum behavior of atoms
- Display tunable absorption and emission spectrum over visible wavelength
- Strong potential for medical imaging
- CdSe core with ZnS Shell



# Biological Effects

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- Ultra small size
  - Easy penetration
  - Fast distribution
  - Direct interaction with organelles
- Huge surface areas
  - Increased chemical activities (catalyst)
  - Increased biological activities

# Two-Sided Sword

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- Beneficial uses
  - Diagnostic (imaging, sensor)
  - Therapeutic (drug delivery)
  - Biomedical research
- Adverse health effects
  - Rapid uptake thru skin & epithelial cells
  - Translocation along neurons
  - Novel or amplified toxicity
  - Oxidative stress

# Preliminary Toxicological Data

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- Buckyballs caused lipid peroxidation in fish brain at 500 ppb for 48 hr
- Nanosize  $\text{TiO}_2$  toxicity increased, proportional to surface area
- SWNT causes oxidative stress: free radical formation, depletion of antioxidants
- Metal impurity may also add toxicity
- Physical dimensions of some NT may cause pulmonary fibrosis and cancer similar to asbestos

# A Series of Concerns

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- Research safety
- Occupational safety
- Consumer safety
- Environmental impact



# Nanoresearch at HKUST

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# Fullerenes Generation by Arc Discharge



Reaction Chamber



Graphite Rods and Powder

# Purification of Fullerenes



# Derivatization of Nanofiber



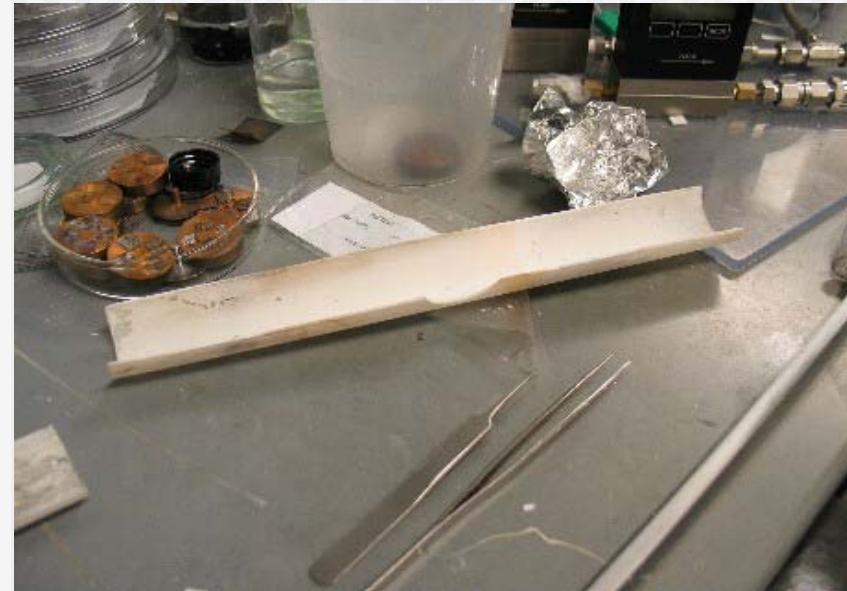
# Carbon Nanotube by Chemical Vapor Deposition





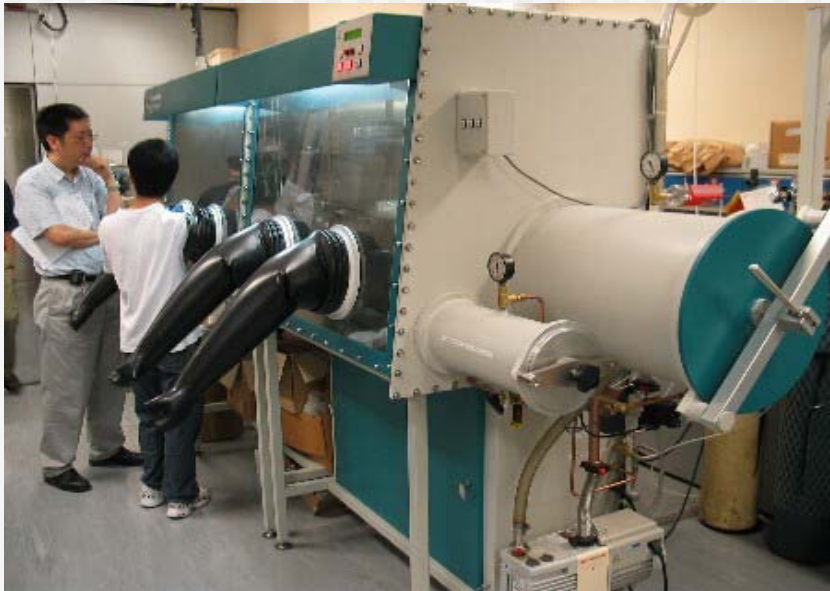
# Nano-sized Zinc Oxide

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# Iron Platinum Nanoparticles

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# Synthetic Zeolite

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- Hydrated alumino-silicate, porous material
- Both natural and synthetic
- Porous structure allow interesting chemistry
- Synthesis can be fine-tuned to produced desired nanostructure
- Potential applications as catalyst, fuel cell etc



# Synthetic Zeolite



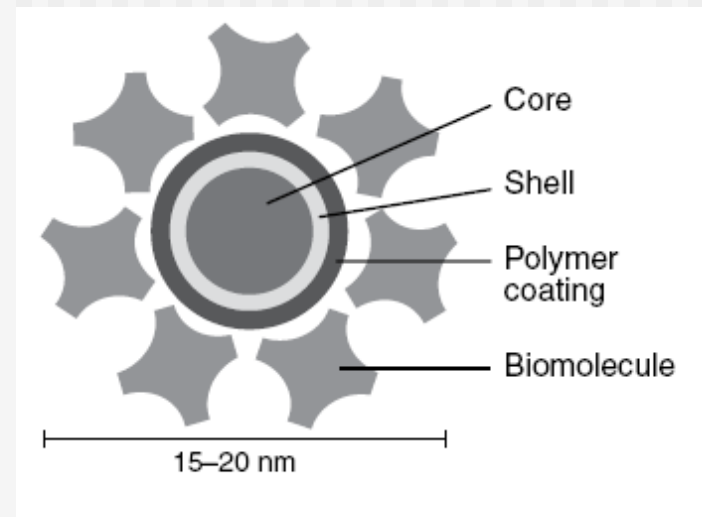
# Other Nanomaterials

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- Nano-Composites
  - Nanosized material e.g.  $\text{CaCO}_3$  added to polymer as filler
  - Carbon nanofiber added to polymer to increase strength
- Silicon Nanowire
  - Silicon nanowire grown on substrate
  - Process under vacuum

# Quantum Dots

- Quantum Dots conjugate used as fluorescence label
- *Xenopus* muscle cells grown on circular coverslip
- Advantage: Qdot does not “bleach” over time like traditional fluorescence



<http://probes.invitrogen.com/products/qdot/overview.html>

# Elements of Nanosafety Program at HKUST

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# Nano, or Otherwise...

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# Scope of Nanosafety Program

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- Nanoparticles ( $d < 100$  nm) & nanomaterials with one extended dimension, e.g. nanotube, nanowire, nanofiber
- Nanocomposite, e.g.  $\text{CaCO}_3$ /PP composite
- Materials with engineered nanostructure, e.g. zeolite with designed nano-size pores
- Silicon-based nanostructure, nanoelectronics
- Nano-enabled products
- “Nano-stuff”

# Preliminary Nanosafety Program at HKUST

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- Safety input to Nanolab facility
- Nanosafety in research proposal review
- Nanomaterial database
- Assessment of exposure potentials
- Nanosafety training and chapter in Safety Manual
- Precautionary measures
- Monitoring of nano-EHS development

# Precautionary Measures for HKUST Researchers

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- Mainly for engineered nanoparticles
- Key references
  - *Approaches to Safe Nanotechnology: An Information Exchange with NIOSH*. National Institute for Occupational Safety and Health, July 2006.  
[http://www.cdc.gov/niosh/topics/nanotech/safenano/pdfs/approaches\\_to\\_safe\\_nanotechnology\\_28november2006\\_updated.pdf](http://www.cdc.gov/niosh/topics/nanotech/safenano/pdfs/approaches_to_safe_nanotechnology_28november2006_updated.pdf)
  - *Nanoscience and nanotechnologies: opportunities and uncertainties*. The Royal Society & Royal Academy of Engineering, July 2004.  
<http://www.nanotec.org.uk/finalReport.htm>



# Precautionary Measures—Risk Assessment

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- Nanomaterials possess unique physical chemical properties, should be considered as “new chemicals”
- Treat as potentially hazardous
- MSDS of raw material can only be viewed as a starting point at best
- Carbon: practically non-toxic; buckyballs: oral LD<sub>50</sub> 300 mg/kg

# Precautionary Measures— Airborne Particles (1)

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- Avoid aerosol generation
  - Many NPs generated in labs under contained conditions
  - NPs tend to agglomerate
  - However, many attempts to disperse NPs for various applications
- Suspended fine powder can become a fire and explosion hazard
- Do not allow escape into environment
- Intentional release into environment must show benefits clearly outweighing risks

# Precautionary Measures— Airborne Particles (2)

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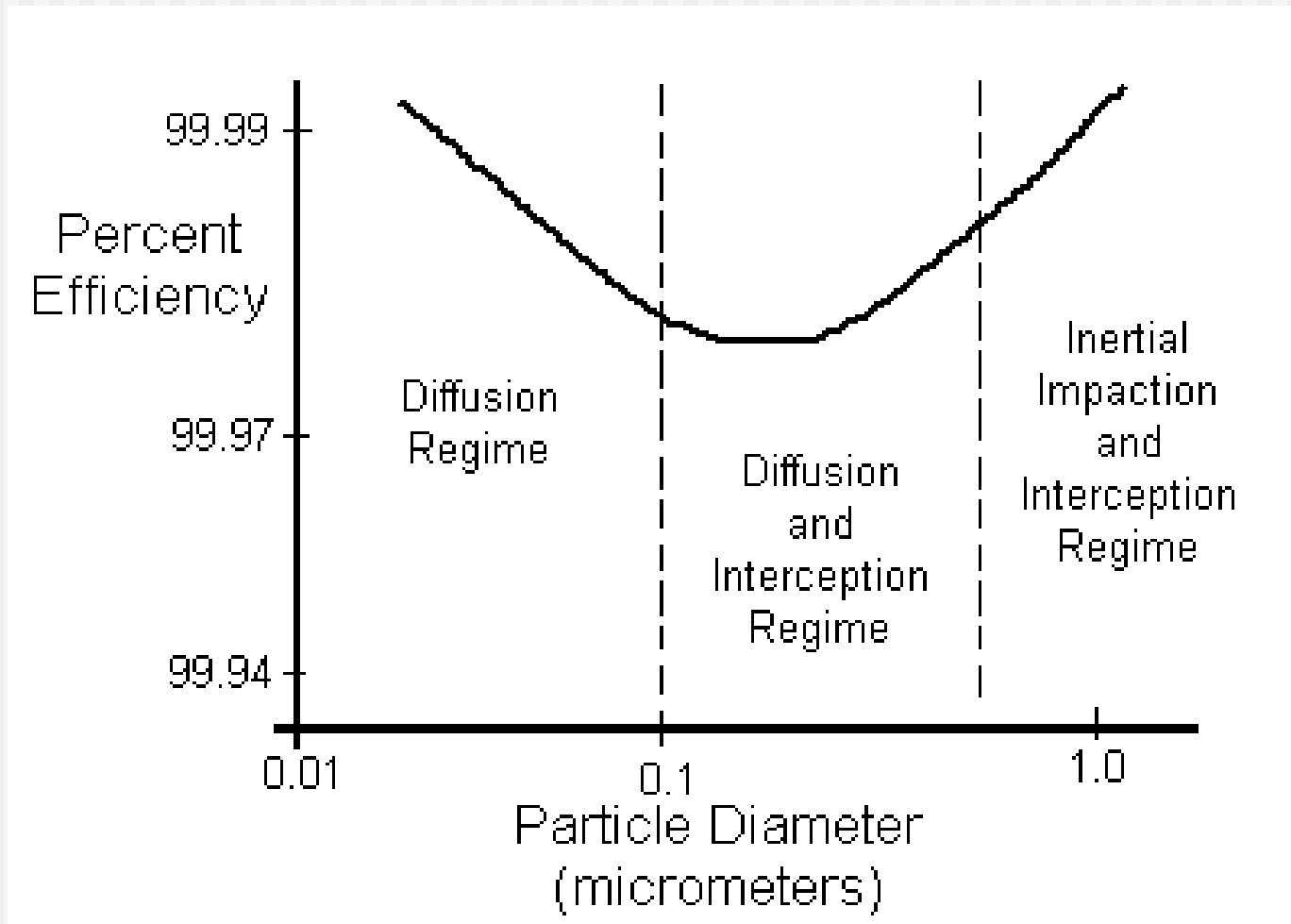
- Use local exhaust ventilation
  - Fumehood
  - Biological safety cabinets
  - Pharmaceutical hoods
- Use HEPA
  - Air treatment
  - Protection of personnel

# Can HEPA Filter Capture Nanoparticles?

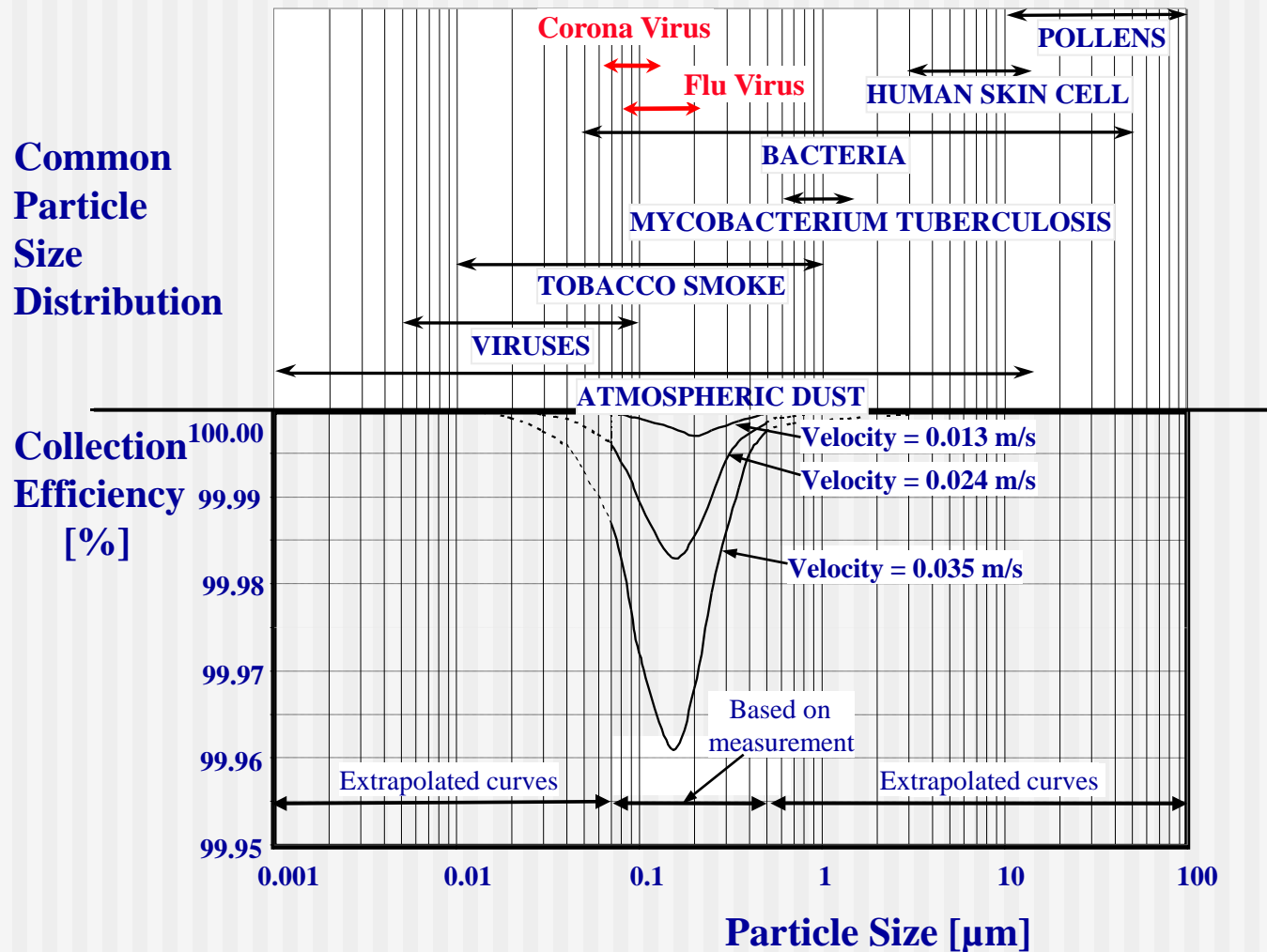
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- Theoretically efficiency increase with decreasing size below 0.1-0.3  $\mu\text{m}$
- Previously confirmed by data point down to 0.01  $\mu\text{m}$  (10 nm)
- Recent study by U Minn confirmed HEPA filtration efficiency down to 2-4 nm, no thermal rebound observed
- Nanoparticles tend to agglomerate, instead of existing as single particles

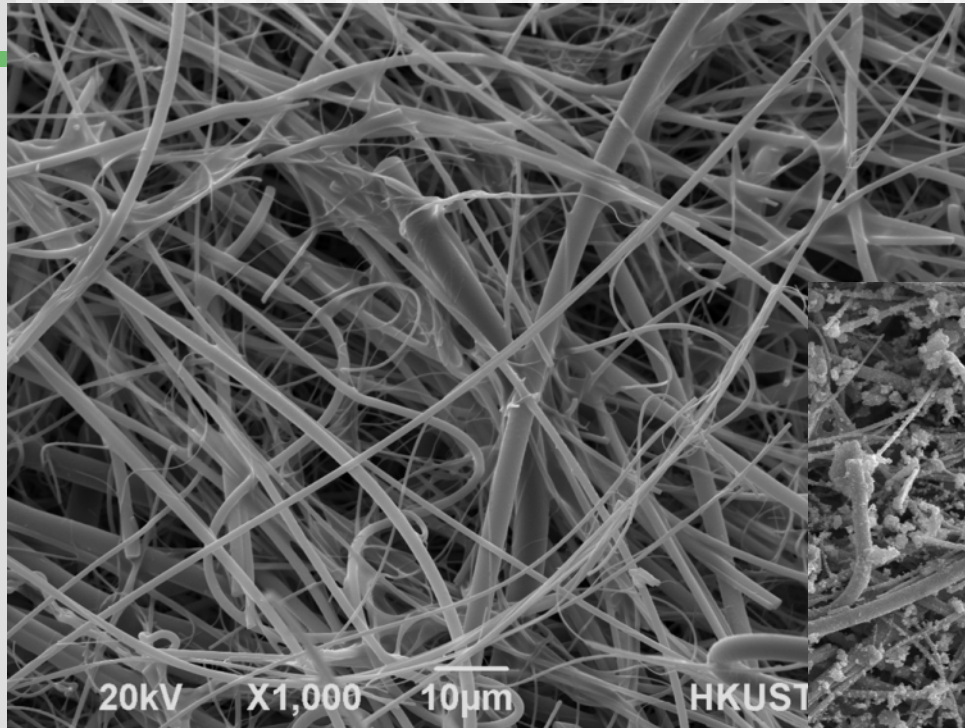
# Particle Size and Filtration Efficiency



# HEPA Filtration Efficiency

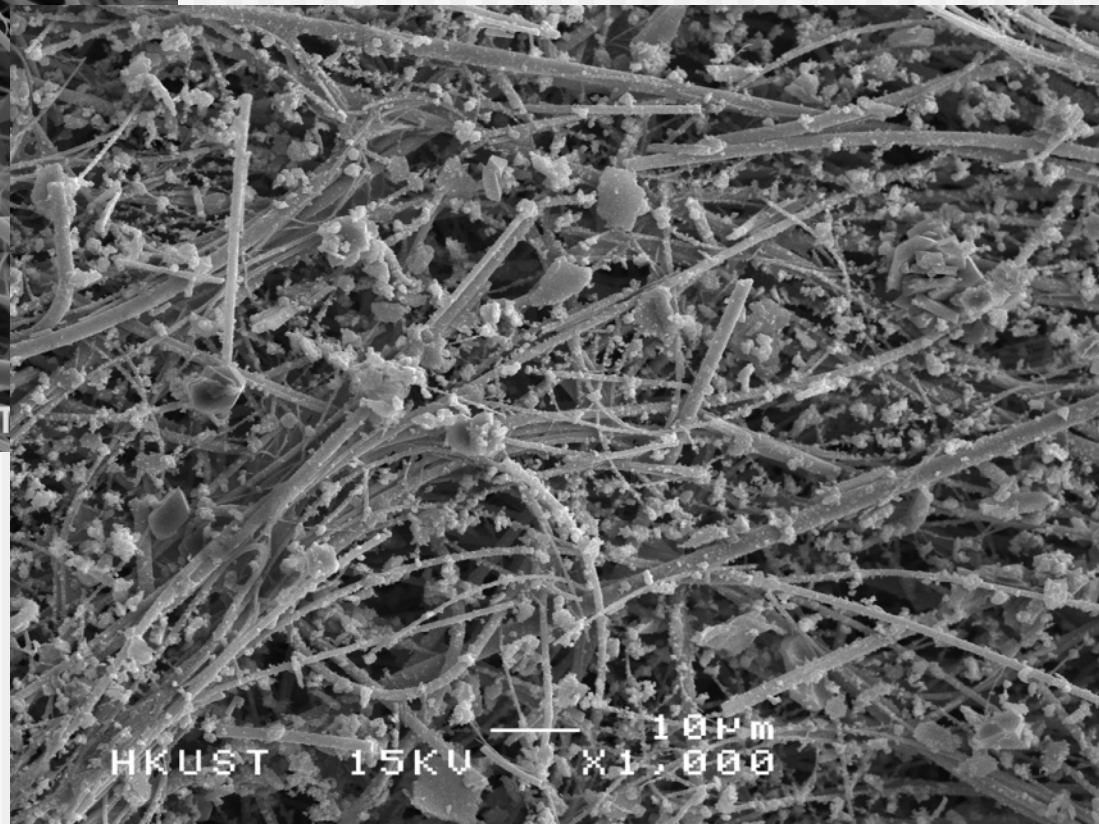


# SEM of Fibrous Filter 1000 x



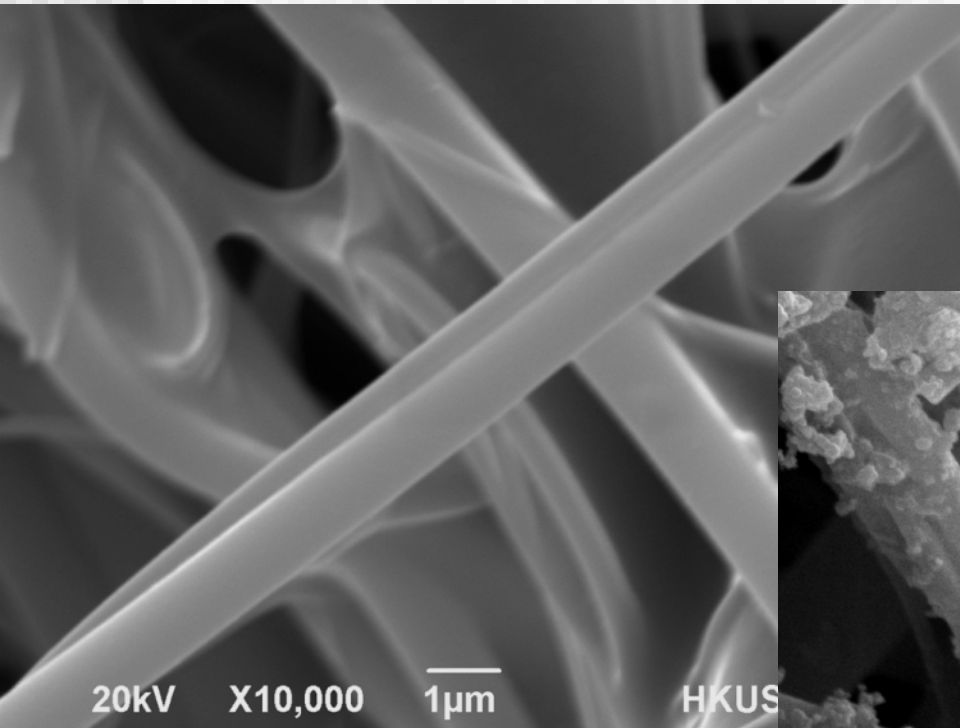
↑  
New Filter

Used Filter





# SEM of Fibrous Filter 10000 x



New Filter



Used Filter



# Precautionary Measures— Liquid Solution/Suspension

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- Avoid aerosol generation
- Avoid skin contact
- Use gloves at least protective against carrier solvent
- Do not allow escape into environment
- Intentional release into environment must show benefits clearly outweighing risks

# Precautionary Measures— Waste Disposal

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- Treat as chemical waste as a minimum
- Seek to remove from waste stream or destruct nanofeature as far as practicable

# Quantum Dots Toxicity

- “We have not investigated the toxicity of the Qdot® streptavidin conjugate. The materials are provided in a solution which is ~2 mM total Cd concentration; however, the CdSe core is encapsulated in a shell of ZnS and the polymer shell, which may prevent dissolution of free Cd. We have demonstrated the utility of these materials in a variety of live-cell in vitro labeling experiments, but do not have systematic data investigating the toxicity of the materials to humans, to animals, or to cells in culture.”

# Quantum Dots Disposal

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- “The Qdot® conjugate contains cadmium and selenium in an inorganic crystalline form. Please dispose of the material in compliance with all applicable local, state, and federal regulations for disposal of these classes of material. For more information on the composition of these materials, consult the Material Safety Data Sheet.”

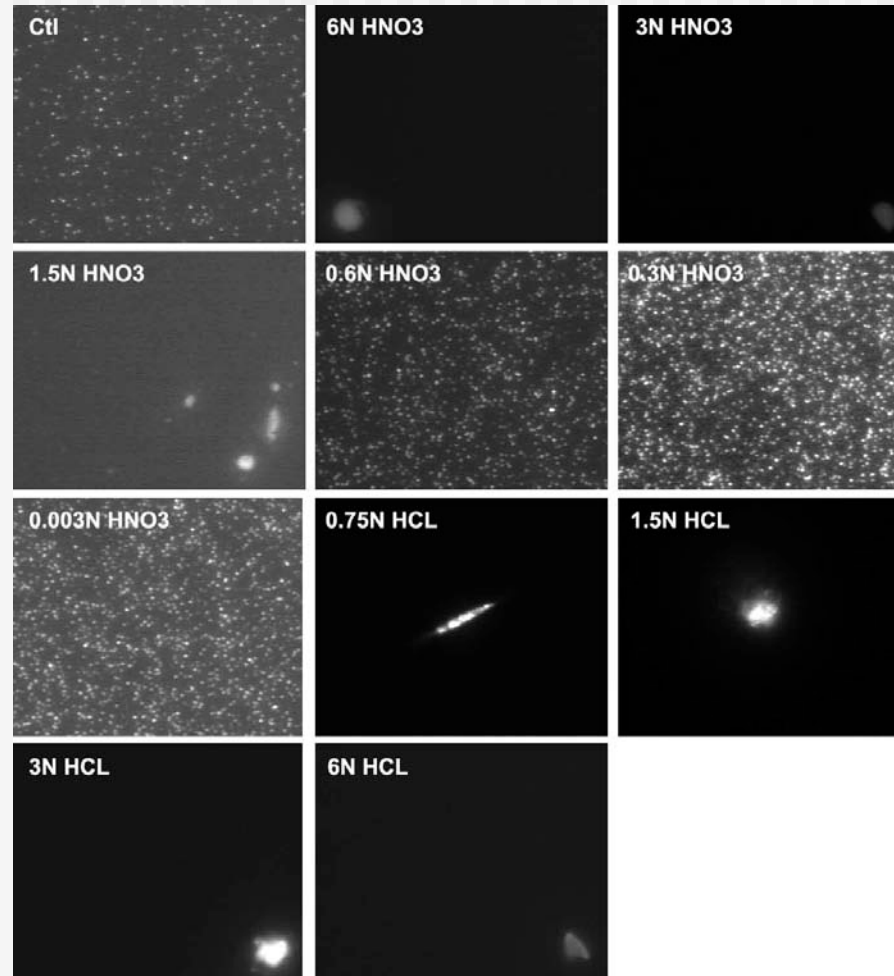
Source: Qdot Streptavidin Conjugates User Manual, Intrivogen

# Qdot Destruction Experiment

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- Tried different concentrations of HCl and  $\text{HNO}_3$  to dissolve Qdots
- Use fluorescence wavelength under microscope to verify destruction
- May need to use TEM as confirmation
- Disposed as waste metal solutions may not destruct nano-feature

# Acid-Treated QDot Solution



# Precautionary Measures— Minor Spill Response

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- Treat as hazardous material spill
- Evacuate area, wait 20 minutes to allow aerosols to settle
- Don respirator, gloves and coverall
- Remove spilled dry nanomaterials by HEPA vacuum cleaner
- Use normal sorbent materials for spilled liquid containing nanomaterials
- Contact HSEO for major spill

# Legal Framework for Chemical Wastes

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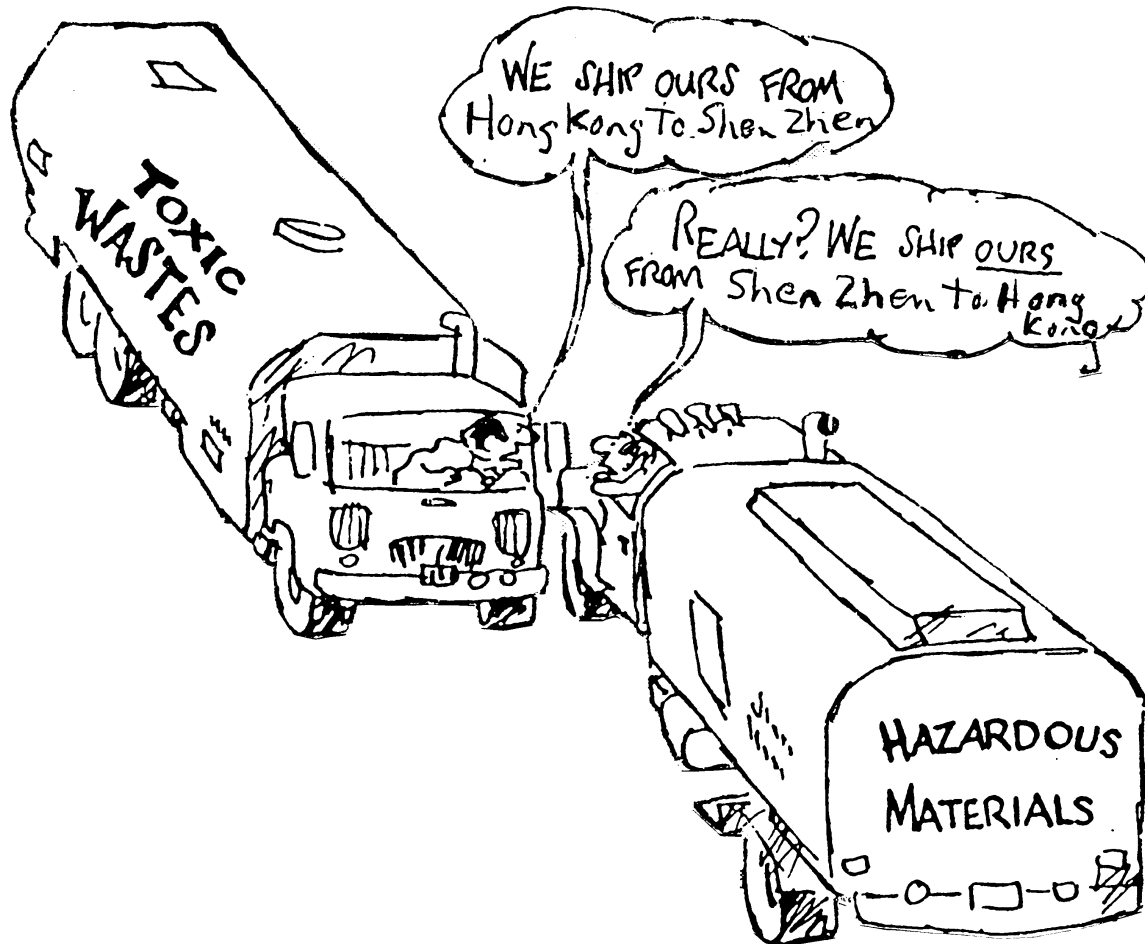


# What Is Hazardous Waste?

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- Chemical wastes
- Biological/infectious wastes
- Medical/clinical wastes
- Radioactive wastes
- Co-contaminated wastes

# A Wrong Way to Deal with Hazardous Waste



# Chemical Waste Regulations in HK

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- Waste Disposal (Chemical Waste) (General) Regulations of 1992
- Set up a waste treatment facility
- Prohibition of disposal into sewer
- Registration of waste generators
- Packaging, labeling and storage of chemical waste
- Documented collection and disposal scheme

# Penalties for Chemical Waste Violations (1)

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<b>Section of the Regulation</b>	<b>Offence</b>	<b>Maximum Penalty</b>
6	Failing to register as a waste producer	\$200,000 plus 6 months imprisonment
8	Failing to arrange for proper disposal of waste	\$200,000 plus 6 months imprisonment
9-11, 13-15	Failure to comply with waste packaging & storage requirements	\$100,000 plus 6 months imprisonment

# Penalties for Chemical Waste Violations (2)

Section of the Regulation	Offence	Maximum Penalty
12, 18	Failure to comply with labeling and warning requirements	\$50,000 plus 6 months imprisonment
37	Furnishing false information	\$200,000 plus 6 months imprisonment
17	Failure to notify in relation to "Part A" waste or to comply with any given directions	1st offence - \$50,000 2nd offence - 100,000 plus \$500/day

# Legal Definition of Chemical Waste

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- Chemical Waste is legally defined by reference to a list of substances and chemicals in Schedule 1 of the Waste Disposal (Chemical Waste) (General) Regulation enacted 18 March, 1992.

# Items Not Considered to be Chemical Waste

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- Dilute acids and alkalines, if % by weight is less than figures in Appendix A of "A Guide to the Chemical Waste Control Scheme", HKEPD, Sep 1992.
- Neutralized salts, not listed in App A.
- Elemental metals, not listed in App A (e.g., Copper, Iron, Tin, Zinc).
- Household chemicals used in labs.
- Apparatus rinsing water or wiping tissue.

# HKUST policy and program

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# HKUST Safety and Environmental Policy

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- ...The University will assume all reasonable directives possible to protect its appointees, students, facilities, and the environment, placing priority emphasis towards...
- ...Proper and safe procedures for the use, handling, storage, transport, and disposal of articles, hazardous materials and waste...

# Chemical Waste Management at HKUST

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**Waste Generators** are responsible to:

- Request the correct waste container from HSEO.
- Safely put their waste into the waste container.
- Fill out the Chemical Waste Log Sheet.

# Chemical Waste Management at HKUST

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**Waste Generators** are responsible to:

- Segregate waste materials in the laboratory.
- Safely and properly store and maintain all waste within the laboratory.
- Request collection of waste by HSEO.

# Chemical Waste Management at HKUST

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**HSEO** is responsible to:

- Provide waste containers and technical assistance to waste generators at HKUST.
- Transport waste from laboratories to the waste accumulation area.
- Manage the HKUST waste accumulation area.

# Chemical Waste Management at HKUST

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**HSEO** is responsible to:

- Prepare hazardous waste for pickup and disposal by the waste contractor.
- Maintain HKUST hazardous waste records.
- Liaise with Environmental Protection Department and Enviropace on hazardous waste issues.

Chemical Waste Collection Annual Summary		
Year	Quantity of Chemical Waste	
	Liquid (L)	Solid (pack)
1994 (Jul-Dec)	2,380	81
1995	10,530	113
1996	14,580	43
1997	20,830	30
1998	28,100	36
1999	29,180	37
2000	28,680	38
2001	31,520	23
2002	41,600	104
2003	38,280	237
2004	40,360	92
2005	33,140	52
2006	42,700	48
2007	35,740	30

# Chemical waste streams

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# Three Main Types

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- Liquid Chemical Waste picked up by Enviropace
- Lab Pack Waste picked up by Enviropace
- Solid Chemical Waste disposal arranged by HSEO



# Enviropace Chemical Waste Streams (1)

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- Halogenated Organic Solvents
- Non-halogenated Organic Solvents
- Organic Acids
- Inorganic Acids
- Hydrofluoric Acid
- Metal Solution
- Chrome (VI) or Boron

# Enviropace Chemical Waste Streams (2)

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- Alkalines
- Cyanides
- Lubricating Oil
- Photographic Film Fixer/Developer
- Gel Waste
- Phenol Sludge
- Metal Sludge
- Rags with Solvent and Oil

# Chemical Waste Mixtures

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- Select the container according to the MAJOR component of the mixture.
- Exceptions:
  - Always put halogenated compounds in the "Halogenated Solvents Container"
  - Always put cyanide waste into the "Cyanide" container (Always maintain alkaline,  $\text{pH} > 7$ , to avoid HCN formation) even if these are the minor component in the mixture.

# Different Chemical Waste Containers

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- **Plastic** (Acids and Alkalies, Fixer/Developer)
- **Carbon Steel** (Lube Oil, Other Oils)
- **Metal with inner plastic lining** (Halogenated & Non-halogenated Solvents)
- **Plastic, Open top** (organic gel)









# Required Information on Waste Containers

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- Waste generator name and phone number
- Waste stream name
- Standard Waste Label with Hazard Warning Symbol(s)
- A Waste Log Sheet in a plastic folder (for applicable waste streams)



# Hazard Symbols on Chemical Waste Labels

Classification 危險分類	Symbol 符號	Classification 危險分類	Symbol 符號
Explosive 爆炸性	 Black image on Orange Background	Toxic 有毒	
Flammable 易燃	 Black image on Red Background	Harmful 有害	
Oxidizing 助燃	 Black image on Yellow Background	Corrosive 腐蝕性	
Irritant 刺激性		Asbestos 石棉	

# Intermediate Waste Containers

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- Intermediate waste containers must be clearly labeled (waste stream, responsible person, contact)
- Containers that receive multiple sources of waste must have a log
- Same handling procedures, including logging and compatibility testing
- If empty reagent bottles are used, they must be thoroughly cleaned



# Chemical waste procedures

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# Waste Handling Safety Precautions

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- Carefully inspect each container for broken handles or leaks before use.
- Use appropriate personal protective equipment: splash-proof goggles, face shield, gloves, lab coat, respirator, etc.
- Use a fume cupboard for waste with hazardous or odorous vapors.
- Close container tightly after transfer.
- Be careful handling full containers.



# Waste Handling Procedures

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- Always check liquid level before adding new liquids.
- Perform compatibility test before adding new liquids.
- Fill container to no more than 75% of full capacity.
- For large volume transfers, use funnel.
- Fill in the chemical waste log sheet immediately.

# Compatibility Test Procedures (1)

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- Performed **inside** a fume cupboard, by an experienced person.
- Fume cupboard must be functioning properly.
- Transfer 50 mL existing waste from target container to beaker.
- Place thermometer into beaker.
- Slowly mix in new chemical waste, to the estimated volumetric ratio.

# Compatibility Test Procedures (2)

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- If bubbling, fuming, or a temperature increase of  $>10^{\circ}\text{C}$  occurs, stop mixing, the wastes are INCOMPATIBLE.
- If no observable reaction occurs within 5 minutes, the wastes are COMPATIBLE.

# Chemical Waste Log Sheet

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- Use a dark color permanent ink pen (not pencil, or water-based ink).
- Use chemical names of reagents and waste (full name in block letters, not abbreviations).
- Include the volume of waste being transferred.
- Estimate the percentage of mixture components.

# The Hong Kong University of Science & Technology

## Chemical Waste Log Sheet

C.W. P. Reg. No.: 9311-831-T2006-01

SEPO No.: \_\_\_\_\_

Waste Type: \_\_\_\_\_

CWTF-ID: \_\_\_\_\_

Department: \_\_\_\_\_ Room No.: \_\_\_\_\_ Contact Person: \_\_\_\_\_ Ext: \_\_\_\_\_

Date (DD/MM/YY)	Name of Reactant/ Product/ Waste (Full Name in Block Letter)	Quantity	Producer's Name	Phone No.



# Why is Chemical Waste Log Sheet Important?

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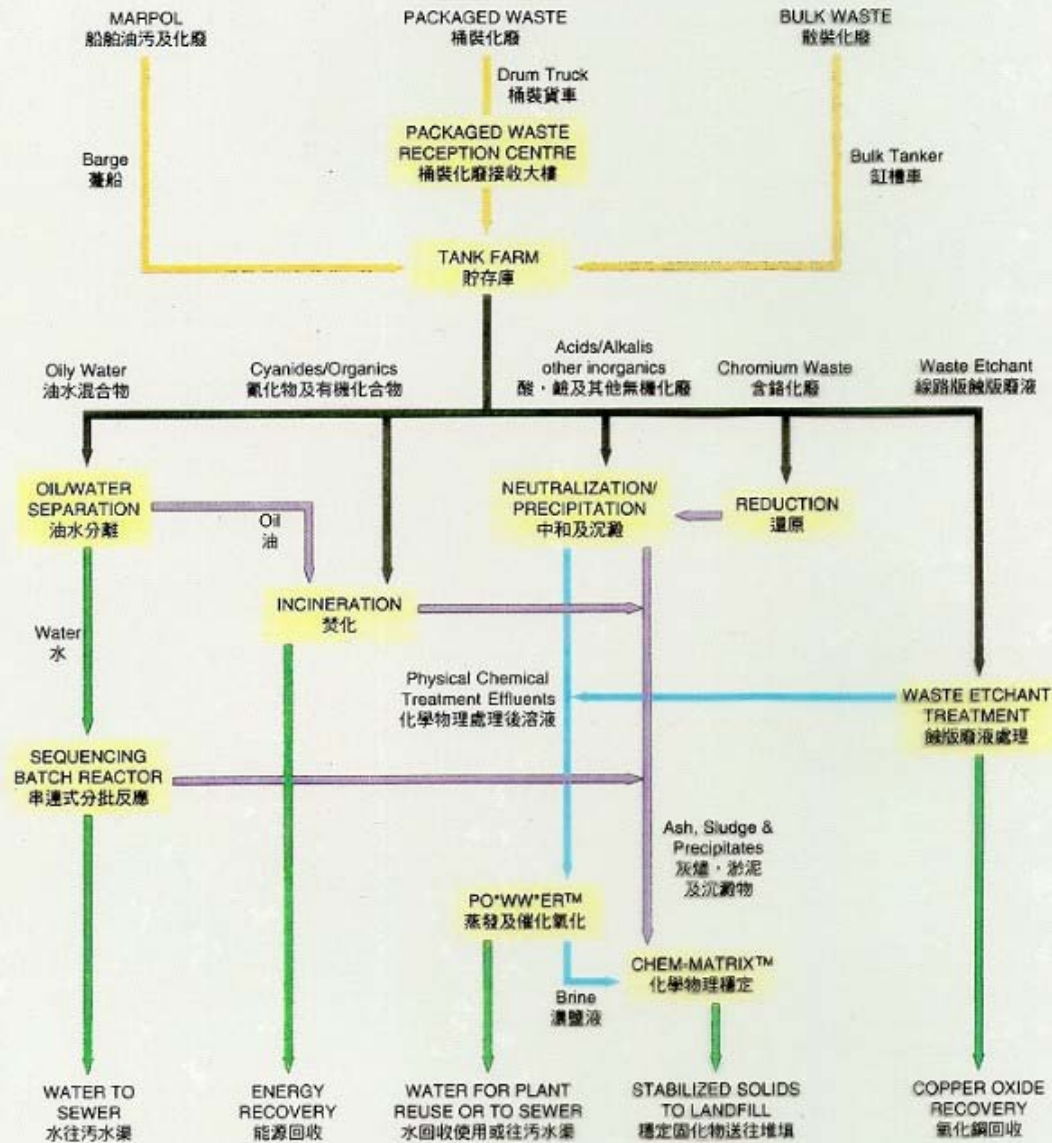
- Requirement of Enviropace: "Every waste entering into a chemical waste container must be properly logged. The filled log sheet must accompany the container at the time of collection by Enviropace. Enviropace will refuse collection of containers that do not carry a legible log sheet."

# Why is Chemical Waste Log Sheet Important?

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- To comply with the “Code of Practice on the Packing, Labelling and Storage of Chemical Wastes” issued by the HK Environmental Protection Department.
- To provide information to Enviropace on the treatment of received waste containers.
- To provide clues about the compatibility of chemical wastes in container.

# CWTC OPERATION FLOW DIAGRAM 化學廢物處理流程



# Chemical Waste Storage

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- Use a spill tray / secondary container for spill containment.
- Keep the waste log sheet current.
- Every storage area must have a proper chemical waste sign.
- Store flammable waste in a flammable cabinet wherever possible.





# Chemical Waste Collection Procedures

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- Use the HSEO web page, “Request for Chemical Waste Collection”.  
<http://www.ab.ust.hk/hseo/sub-menu/waste.htm>
- Waste collection service by HSEO is on Tuesday, or the next working day if Tuesday is a holiday.
- Users should make a pickup appointment if a collection site is unattended.

# Lab-Pack Program

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- Highly reactive chemicals, water reactive chemicals, including conc. strong oxidizing or reducing agents
- Waste that cannot pass the compatibility test
- Unwanted or expired chemicals
- Disposed of in original containers
- Valuable for labs, unfortunately program not running well due to cost



# Current Lab Pack Practice

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- Generator produce list of chemicals need to be disposed
- EPD/Enviropace review list then give advice:
  - Dissolve/dilute chemical then put in suitable liquid chemical waste stream
  - Put originally packaged chemical in specified standard waste container
  - Enviropace pick up in regular rounds

**The Hong Kong University of Science and Technology**  
**Safety and Environmental Protection Office**  
**Labpack Waste (Unwanted / Expired Chemical) Form**

Name of waste producer: Leo LEE Dept: AEMF Post: Senior Technical Officer  
 Tel: 8665 Room: 4230 E-mail: chlee@ust.hk  
 Name of supervisor: \_\_\_\_\_ Post: \_\_\_\_\_ Tel: \_\_\_\_\_ E-mail: \_\_\_\_\_  
 Waste location: 4229  
 Date (DD/MM/YY): 27/08/09

Chemical Name	Chemical Formula	Physical State (S/L/M)*	Container Size (ml)#	Container Type (G/P/M)^	Lid Type (G/P/M)	No. of Container	Total amount (Kg/Litre)	Remarks (e.g. safety precaution, condition of container)
Potassium Permanganate	KMnO <sub>4</sub>	S	500	G	P	1	500g	
Potassium Permanganate	KMnO <sub>4</sub>	S	1000	M	M	1	1 kg	

\* S = solid, L = liquid, M = sludge

# size of container holding chemical

^ G = glass, P = plastic, M = metal

Note (SEPO use only): \_\_\_\_\_



**COMPLETED**

COMPANY NAME : THE HONG KONG UNIVERSITY OF SCIENCE & TECHNOLOGY

CWPN : 9311-831-T2006-01

WPS1 DATE : 11 September 2009

TOTAL ITEMS : 4

CHEMICAL ITEMS (PRODUCER TO NOTE / FURTHER ACTION)

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REJECT DUE TO : N/A

PROVIDE MSDS OR CHEMICAL COMPOSITION : N/A

CHECK NAME : N/A

CHECK FORMULA : N/A

CHECK pH : N/A

CHECK PHYSICAL STATE : N/A

CHECK CONTAINER TYPE : N/A

EPD "PART A" WASTE : N/A

C&E "CONTROL SUBSTANCE" WASTE : N/A

**COMPLETED**

## Labpack Waste Classification List - Matched

Waste Name	Waste Ctnr Code	Code	Color	Pre-Treatment	CWTFID	Item	Remark
UNWANTED BROMINE, LABPACK	L36	I x 1	GOLD6	MAKE SURE THE LID IS TIGHTLY ENCLOSED AND THE BOTTLE IS NO LEAKING. WRAP EACH BOTTLE WITH PLASTIC BAG AND PACK IN A CARTON BOX. PUT THE CARTON BOX INTO OUR 20-L PLASTIC PAIL FOR RECEIVING.	10247548	4	
SPENT OXIDIZER, ACIDIC, LAB	L48	A x 24	GRAY	DISSOLVE & DILUTE W/ WATER INTO ~5% (i.e.-WASTE:WATER=1:20) AND THEN TRANSFER INTO OUR PAIL.	10195577	1	
SPENT OXIDIZER, ALKALI, LAB	L58	A x 2	GRAY4	DISSOLVE & DILUTE W/ WATER INTO ~5% (i.e.-WASTE:WATER=1:20) AND THEN TRANSFER INTO OUR PAIL.	10195578	2 3	

*tracking form*

↑

ITEMS MATCHED : 4

## CONTAINER CODE :

A = 20L PLASTIC (CLOSED TOP) PAIL  
 B = 20L CARBON STEEL PAIL  
 I = 20L PLASTIC (OPEN TOP) PAIL  
 \* = 20L PLASTIC PAIL IN 30L CARBON STEEL PAIL  
 J = 50L FIBRE DRUM  
 K = 100L FIBRE DRUM  
 P = PRODUCER OWN PAIL (<= 50L)  
 M = PRODUCER OWN PAIL (> 50L)

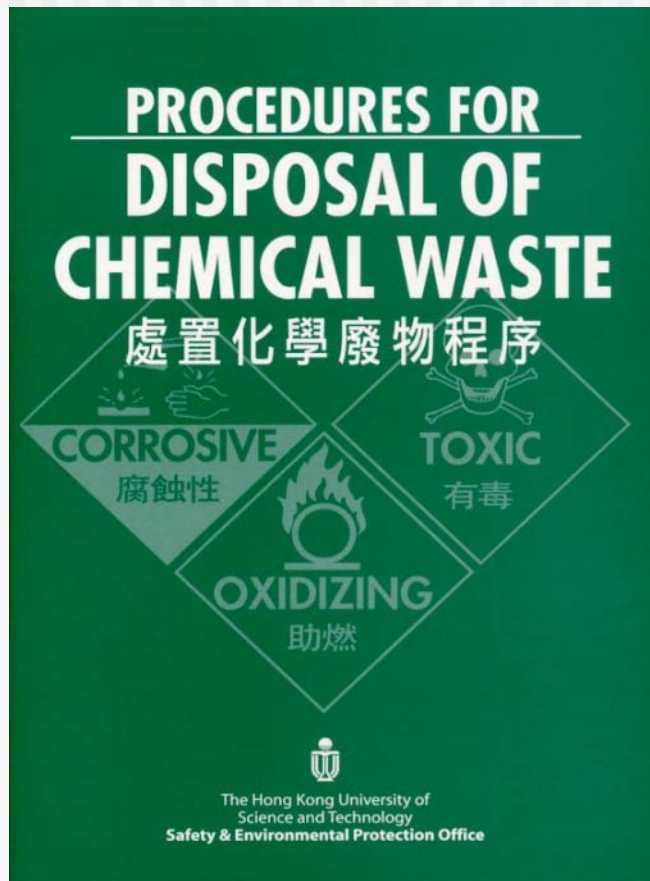
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# Solid Chemical Wastes

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- Elemental mercury waste, periodically collected by Enviropace (Don't use mercury thermometers!)
- Other solid chemical wastes, such as labware in contact with highly hazardous chemicals
- Special disposal arrangement through HSEO

# Information on Chemical Waste Disposal



- Chemical waste lists
- Flowcharts to determine waste type
- Waste handling procedures
- Bilingual text
- [http://www.ab.ust.hk/hseo/sub-menu/chem\\_waste.htm](http://www.ab.ust.hk/hseo/sub-menu/chem_waste.htm)

# Incidents & lessons learned

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# Case 1: Serious Injury

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- Lab user wanted to dispose of several mL of conc nitric acid
- An empty reagent bottle intended to be used as intermediate container for non-halogenated solvent was used
- An explosion occurred several minutes after transfer and closing of the cap
- Serious injury to one hand of the user requiring surgery and hospitalization







# Case 2: Serious Damage

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- IC is regularly etched with conc acid, then wash with excessive solvent
- Mixed waste collected in a non-halogenated solvent
- Lab user added a large quantity of used acid into same container
- Container exploded causing extensive damage to cabinet and rest of the lab











# Case 3: Uncontrolled Release

---

- Lab user macerate silicon plastic with conc sulfuric acid
- Waste was put into spent acid container with other acids including nitric
- Fume was observed, lab user capped the container and left
- Container later ruptured shooting waste onto ceiling and surrounding areas



# Lessons Learned

---

- Oxidizing (e.g. nitric) acid mixed with organics is the most common cause of chemical waste incidents
- Clean intermediate waste container thoroughly, if not sure, clean again
- Plan beforehand, especially for non-routine operation, may need to initiate a new waste stream

# Lessons Learned

---

- Dilute conc acid (10 times) before transfer into waste container
- If fuming is observed after waste transfer, move the container into a running fume hood or open area
- NEVER CAP container with ongoing gas generating reaction inside
- Perform compatability test, NEVER mix incompatible waste!

# Waste minimization

---

# Waste Minimization

---

Users are encouraged to:

- Estimate usage rate carefully.
- Buy in container sizes appropriate to actual use. (Money savings from bulk purchases are easily consumed by the cost of waste disposal.)
- Reduce inventory to a minimum:  
(rotate old containers to front of shelf, review stock before purchase, etc.).

# Waste Minimization

---

- Waste disposal costs are increasing. Recharge to waste generators in Hong Kong is current practice.
- Consider trading chemicals with other researchers before disposal.

# Chemical Exchange Program

---

- Chemicals left after completion of projects
- Some are high quality chemicals in factory seals
- Give away free of charge
- Details in HSEO web page "Chemical Exchange Program"
- [https://www.ab.ust.hk/wkx/kx\\_intro.html](https://www.ab.ust.hk/wkx/kx_intro.html)

# Co-contaminated Wastes

---

- Chemical/Biological
  - Disinfect then treat as chemical waste
- Chemical/Radioactive
  - Treat as radioactive waste
- Biological/Radioactive
  - Chemically disinfect then treat as radioactive waste
- Do not autoclave waste with volatile organics or radioisotopes

# Hazardous Materials Inventory System

---



# Purposes

---

- A “cradle to grave” approach to track the usage of hazardous materials on campus
- A centralized system to assist University management and users to better manage storage and use of hazardous materials
- A platform for centralized purchasing and/or delivery of hazardous materials
- A crucial compliance tool for upcoming amendment of DG regulations

# Features Highlights

---

- Allow different levels of access authority
- Radioisotopes, legally controlled substances, and chemical warfare agents are tracked. Confirmation by lab supervisor is required for transfer or disposal
- Time-sensitive chemicals such as peroxidizables, picric acid, are flagged to remind user for checking
- Allow searching of specific types of chemicals such as DG by classes, other regulated chemicals, which helps regulatory compliance

# Hazardous Material Inventory Management System Overview

HKUST 香港科技大學

## Hazardous Material Inventory Management System

User : istest41

### System Administration

- [User Profile](#)
- [Dept. Profile](#)
- [Lab Profile](#)
- [Hazard Category](#)
- [IMDG Class](#)

### Chemical Transaction

- [Chemical Profile](#)
- [Chemical Direct Purchase](#)
- [Chemical Import](#)
- [Chemical Export](#)
- [Chemical Transfer](#)
- [Chemical Stock Take](#)
- [Chemical Inventory Check](#)

### Alerts

- [Controlled Chemical](#)
- [Peroxide test](#)

### Rad. Material Transaction

- [Rad. Material Profile](#)
- [Rad. Material Direct Purchase](#)
- [Rad. Material Import](#)
- [Rad. Material Export](#)
- [Rad. Material Transfer](#)
- [Rad. Material Stock Take](#)
- [Rad. Material Inventory Check](#)

### Biohazardous Material

- [Biohazardous Material](#)

### Information Page

- [Emergency Guideline](#) PDF
- [Radionuclide Notes](#) PDF
- [Risk Phrases](#) PDF
- [Safety Phrases](#) PDF
- [Useful links](#)

## Hazardous Material Inventory Management System

System Administration | Chemical | Rad. Material | Alert | Information Page | Home

User Profile | Dept. Profile | Lab Profile | Hazard Category | IMDG Class

Dept. Profile Enquiry | New Dept. Profile Form

User : istest41

## Dept. Profile Enquiry

☐ Mark All [Next>>](#)

	Code	Dept. Name	Contact Person	Room No.	Contact Tel. No.	
<input type="checkbox"/>	AEMF	Advanced Engineering Materials Facility	Leo C H LEE	4230	8665	<input type="button" value="Detail"/>
<input type="checkbox"/>	AMCE	Atmospheric, Marine & Coastal Environment Program	Joanne On On LEE	102	8394	<input type="button" value="Detail"/>
<input type="checkbox"/>	APCF	Animal and Plant Care Facility	William K W CHAU	7201	8632	<input type="button" value="Detail"/>
<input type="checkbox"/>	BICH	Biochemistry	Frances Y Y CHUEN	6302	7280	<input type="button" value="Detail"/>
<input type="checkbox"/>	BIOL	Biology	Wai Pang LAM	6214	7327	<input type="button" value="Detail"/>
<input type="checkbox"/>	BRI	Biotechnology Research Institute	Tracy H H PANG	3108A	8970	<input type="button" value="Detail"/>
<input type="checkbox"/>	CAMP	Center for Advanced Microsystems Packaging	Jeffery C C LO	1034A	8356	<input type="button" value="Detail"/>
<input type="checkbox"/>	CENG	Chemical Engineering	Siu Hung LAI	7114	2358 7140	<input type="button" value="Detail"/>
<input type="checkbox"/>	CHEM	Chemistry	Ms Judy W Y TSE	4524	7403	<input type="button" value="Detail"/>
<input type="checkbox"/>	CIVL	Civil Engineering	Kam Shing IU	1203A	7167	<input type="button" value="Detail"/>

Department and laboratory profiles are established during the initial setup of the system

## Hazardous Material Inventory Management System

System Administration | Chemical | Rad. Material | Alert | Information Page | Home

User Profile | Dept. Profile | Lab Profile | Hazard Category | IMDG Class

Dept. Profile Enquiry | New Dept. Profile Form

User : istest41

## Dept. Profile Detail

① Click "Modify" button to edit profile.

<b>Dept. Code</b>	APCF	<b>Room No.</b>	7201
<b>Dept. Name</b>	Animal and Plant Care Facility	<b>Contact Tel. No.</b>	8632
<b>Contact Person</b>	William K W CHAU	<b>Fax No.</b>	--
<b>Contact E-mail</b>	bokwchau@ust.hk		

## Hazardous Material Inventory Management System

System Administration | Chemical | Rad. Material | Alert | Information Page | Home

User Profile | Dept. Profile | Lab Profile | Hazard Category | IMDG Class

Lab Profile Enquiry | New Lab Profile Form

User : istest41

## Lab Profile Enquiry

Room No.	Department	Building	UST Zone	Floor	
<input type="text"/>	CLS <input type="button" value="v"/>	All <input type="button" value="v"/>	All <input type="button" value="v"/>	<input type="text"/>	<input type="button" value="Search"/>

☐ Mark All [Next>>](#)

	Room No.	Building	UST Zone	Floor	Contact Person	Contact Tel. No.	
<input type="checkbox"/>	1036(CLS)	Main building	J	1	Painchai	6883	<input type="button" value="Detail"/>
<input type="checkbox"/>	1037(CLS)	Main building	J	1	Painchai	6883	<input type="button" value="Detail"/>
<input type="checkbox"/>	1212(CLS)	Main building	L	1	Paul K H CHAN	6884	<input type="button" value="Detail"/>
<input type="checkbox"/>	1215(CLS)	Main building	J	1	Painchai	6883	<input type="button" value="Detail"/>
<input type="checkbox"/>	1221(CLS)	Main building	J	1	Painchai	6883	<input type="button" value="Detail"/>
<input type="checkbox"/>	1222(CLS)	Main building	J	1	Painchai	6883	<input type="button" value="Detail"/>
<input type="checkbox"/>	1223(CLS)	Main building	J	1	Painchai	6883	<input type="button" value="Detail"/>
<input type="checkbox"/>	1227(CLS)	Main building	J	1	Painchai	6883	<input type="button" value="Detail"/>

## Individual laboratory profile

## Hazardous Material Inventory Management System

System Administration | Chemical | Rad. Material | Alert | Information Page | Home

User Profile | Dept. Profile | Lab Profile | Hazard Category | IMDG Class

Lab Profile Enquiry | New Lab Profile Form

User : istest41

## Lab Profile Detail

Click "Modify" button to edit profile.

<b>Room No.</b>	1036	<b>Department</b>	CLS
<b>UST Zone</b>	J	<b>Floor</b>	1
<b>Fax No.</b>	--	<b>Building</b>	Main building
<b>Contact Person</b>		<b>Contact Tel. No.</b>	<b>E-mail Address</b>
1 Painchai	6883		Ista
2 --	--		--
3 --	--		--

## Hazardous Material Inventory Management System

System Administration | Chemical | Rad. Material | Alert | Information Page | Home

User Profile | Dept. Profile | Lab Profile | Hazard Category | IMDG Class

Hazard Category Enquiry | New Hazard Category Form

User : istest41

## Hazard Category Enquiry

☐ Mark All [Next>>](#)

	Hazard Category	DG Cat. No	IMDG Class	Properties	Remark	
<input type="checkbox"/>	Carcinogens	N/A	N/A	--	11th Report on Carcinogens, NTP and IARC	<input type="button" value="Detail"/>
<input type="checkbox"/>	Chemicals under Dangerous Drugs Ordinance	N/A	N/A	--	Dangerous Drugs Ordinance (Cap 134)	<input type="button" value="Detail"/>
<input type="checkbox"/>	Chemicals under the Control of Chemicals Ordinance	N/A	N/A	--	Control of Chemicals Ordinance (Cap 145)	<input type="button" value="Detail"/>
<input type="checkbox"/>	Corrosives	3	8	Corrosive substances: means substances which, by chemical action, will cause severe damage when in contact with living tissue or, in the case of leakage, will materially damage, or even destroy, other goods or the means of transport.	--	<input type="button" value="Detail"/>
<input type="checkbox"/>	Cyanide containing chemicals	4	6.1	Toxic substances: These are substances liable either to cause death or serious injury or to harm human health if swallowed or	--	<input type="button" value="Detail"/>

Chemicals are classified under several hazard categories and the International Maritime Dangerous Goods Code (IMDG) in the inventory system.

IMDG has been adopted as the basis of the Dangerous Goods classification of the revised DG Ordinance in Hong Kong.

## Hazardous Material Inventory Management System

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User Profile | Dept. Profile | Lab Profile | Hazard Category | IMDG Class

IMDG Class Enquiry | New IMDG Class Form

User : istest41

## IMDG Class Enquiry

☐ Mark All [Next>>](#)

	IMDG Class	DG Cat. No	Properties	
<input type="checkbox"/>	1	1	Explosives	<input type="button" value="Detail"/>
<input type="checkbox"/>	2.1	2	Gases: Flammable gases	<input type="button" value="Detail"/>
<input type="checkbox"/>	2.2	2	Gases: Non-flammable, non-toxic gases Gases which are transported at a pressure not less than 280 kPa at 20°C, or as refrigerated liquids, and which: 1: are asphyxiant - gases which dilute or replace the oxygen normally in the atmosphere; or 2: are oxidizing - gases which may, generally by providing oxygen, cause or contribute to the combustion of other material more than air does; or 3: do not come under the other classes.	<input type="button" value="Detail"/>
<input type="checkbox"/>	2.3	2	Gases: Toxic gases Gases which: 1: are known to be so toxic or corrosive to humans as to pose a hazard to health; or 2: are presumed to be toxic or corrosive to humans because they have a LC50 value (as defined in 2.6.2.1 of IMDG Code 2000 Edition) equal to or less than 5,000 ml/m <sup>3</sup> (ppm).	<input type="button" value="Detail"/>
<input type="checkbox"/>	3	5	Flammable liquids Class 3 includes Flammable liquids and Liquid desensitized explosives	<input type="button" value="Detail"/>
<input type="checkbox"/>	4	6, 8, 9	Flammable solids; substances liable to spontaneous combustion; substances which, in contact with water, emit flammable gases. Class 4 deals with substances (other than those classified as explosives) which, under conditions of transport, are readily combustible or may cause or contribute to a fire.	<input type="button" value="Detail"/>

## New Direct Purchase Form

① Fill in the information and click "Next"

\* denote compulsory information

Date of receiving  
(dd/mm/yyyy) 26 / 09 / 2006

P.O. No. \* NA

Storage Location \* Please select

Owner \* Panchai

Owner's E-mail \* Istan

Ordered By \* Paul Chan

Order Contact Info. \* 6884

② Report your stock item(s) :

Chemical Name \* (-)-Arterenol, Bitartrate Salt, crystalline Find Chemical

CAS No. 69815-49-2 UN No. N/A

Impurity Concentration 98%

Grade Quantity \* 2

Packing \* 5 g

Brand \* Sigma Product No. NA

Company \* -- Unit Price(HK\$) --

Expiry Date \* -- Manufacturing Date --

The system allows reporting of any chemicals purchased directly by users. Once the record is verified by CLS, the record will be added to the inventory of the storage location and a barcode for the item will be sent to the user.

For all purchase made through CLS, a barcode will be affixed on the item and the inventory of the storage location will be updated automatically.

## New Direct Purchase Form

① Click the "Confirm Add" button if you confirm; otherwise click "Back" button for repetition

Date of receiving	26/09/2006	Storage Location	2122 (SEPO)
P.O. No.	NA	Owner's E-mail	Istan
Owner	Panchai	Order Contact Info.	6884
Ordered By	Paul Chan		

Chemical Name	(-)-Arterenol, Bitartrate Salt, crystalline		
CAS No.	69815-49-2	UN No.	N/A
Impurity	--	Concentration	98%
Grade	--	Quantity	2
Packing	5 g		
Brand	Sigma	Product No.	NA
Company	--	Unit Price(HK\$)	--
Expiry Date	--	Manufacturing Date	--

Confirm Add Back

# Chemical Inventory

HKUST 香港科技大學

## Hazardous Material Inventory Management System

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Direct Purchase

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 Inventory Check

User : istest41

### Inventory Check

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Chemical Name	Item Code	Bar Code	Storage Location	Cabinet	Packing	Status	
1,2-Dichloroethane	5	0609210480	1222(CLS)	--	500 ml	Enabled	<a href="#">Detail</a>
1,2-Dichloroethane	5	0609210479	1222(CLS)	--	500 ml	Enabled	<a href="#">Detail</a>
1,2-Dichloroethane	5	0609210478	1222(CLS)	--	500 ml	Enabled	<a href="#">Detail</a>
1,2-Dichloroethane	5	0609210477	1222(CLS)	--	500 ml	Enabled	<a href="#">Detail</a>
1,2-Dichloroethane	5	0609210476	1222(CLS)	--	500 ml	Enabled	<a href="#">Detail</a>
1,2-Dichloroethane	2883	0609210750	1222(CLS)	--	2 L	Enabled	<a href="#">Detail</a>
1,2-Dichloroethane	2883	0609210749	1222(CLS)	--	2 L	Enabled	<a href="#">Detail</a>
1,2-Dichloroethane	2883	0609210748	1222(CLS)	--	2 L	Enabled	<a href="#">Detail</a>
1,2-Dichloroethane	2883	0609210747	1222(CLS)	--	2 L	Enabled	<a href="#">Detail</a>
1,4-Dioxane	2737	0609222323	2247(CLS)	--	1 L	Enabled	<a href="#">Detail</a>

[<<Prev](#) [Next>>](#)



# Information Page

■ Contains information and links which help users manage chemical safety and procurement

HKUST 香港科技大學

## Hazardous Material Inventory Management System

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User : istest41

### Information Page

- [Emergency Guideline](#) PDF [Radionuclide Notes](#) PDF [Risk Phrases](#) PDF [Saftey Phrases](#) PDF
- **Useful Links**
  - Hong Kong Chemical Control Regulation  
<http://www.ust.hk/~webcls/chemcon.htm>
  - List of chemical restricted by Hong Kong Ordinance  
<http://www.ust.hk/~webcls/chemreq.htm>
  - Regulation of Dangerous Goods in Hong Kong Waters, Marine Department, HKSAR  
[http://sc.info.gov.hk/gb/www.mardep.gov.hk/en/ele\\_services/dgis.html#\\_Definition\\_of\\_Dangerous\\_1](http://sc.info.gov.hk/gb/www.mardep.gov.hk/en/ele_services/dgis.html#_Definition_of_Dangerous_1)
  - Dangerous Goods Ordinance, Chapter 295, Laws of Hong Kong and its sub-leg Regulations, HKSAR  
[http://www.justice.gov.hk/blis\\_export.nsf/CurAllEngDocAgent?OpenAgent&Chapter=295](http://www.justice.gov.hk/blis_export.nsf/CurAllEngDocAgent?OpenAgent&Chapter=295)
  - Scheduled Chemical under the Chemical Weapons (Convention) Ordinance, Cap 578, HKSAR  
[http://www.justice.gov.hk/blis\\_ind.nsf/CurAllEngDoc?OpenView&Start=549&Count=30&Collapse=575.1#575.1](http://www.justice.gov.hk/blis_ind.nsf/CurAllEngDoc?OpenView&Start=549&Count=30&Collapse=575.1#575.1)
  - Hazardous Chemical Information**
    - Shock sensitive chemical*  
[http://www.ab.ust.hk/sepo/chem\\_info/shock-sensitive-chemicals.htm](http://www.ab.ust.hk/sepo/chem_info/shock-sensitive-chemicals.htm)
    - Peroxide forming chemical*  
[http://www.ab.ust.hk/sepo/chem\\_info/peroxide-chemicals.htm](http://www.ab.ust.hk/sepo/chem_info/peroxide-chemicals.htm)
    - Peroxide test*  
[http://www.ab.ust.hk/sepo/chem\\_info/peroxide-test.htm](http://www.ab.ust.hk/sepo/chem_info/peroxide-test.htm)
  - Material Safety Data Sheets**  
[http://www.ab.ust.hk/sepo/chem\\_info/MSDS.htm](http://www.ab.ust.hk/sepo/chem_info/MSDS.htm)
  - HKUST Safety Manual**  
<http://www.ab.ust.hk/sepo/sm97/toc.htm>
  - Center of Laboratory Supplies**
    - List of chemical available in stock*  
<http://www.ust.hk/~webcls/chemstock.htm>
    - Links of local and overseas chemical suppliers*  
<http://www.ust.hk/~webcls/links.htm>

# What Should Users Do?

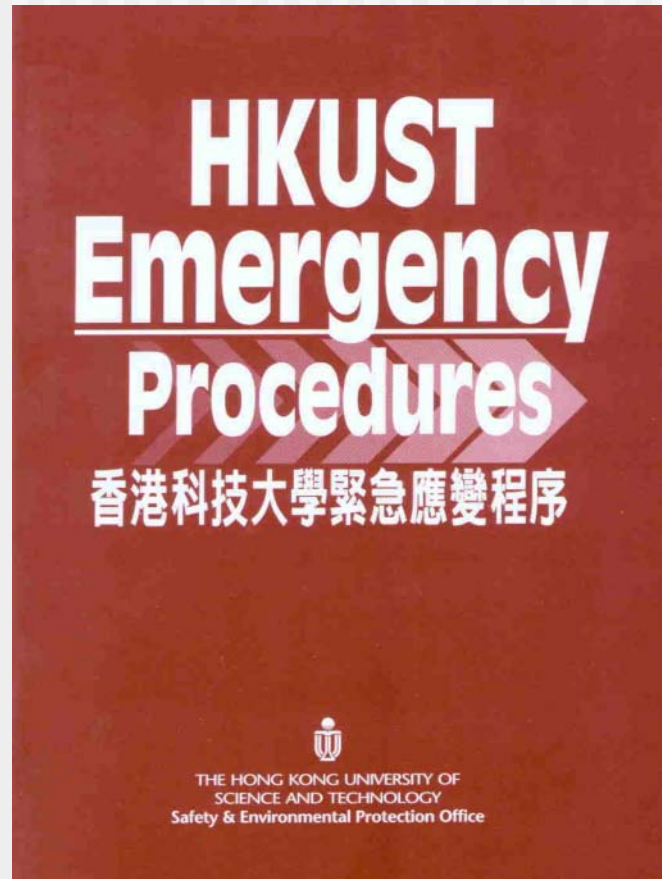
---

- Make sure chemicals not purchased thru CLS are entered into inventory and barcode affixed to container
- Check inventory (own lab and others) before buying chemicals
- Check out empty reagent bottles to keep inventory updated
- Make sure stock-taking is done at least once a year

# Emergency Procedures

---

# Information on Campus Emergency Response

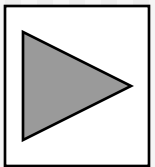


- Important phone numbers
- Emergency preparedness
- Response procedures for various incidents
- Bilingual text
- <http://www.ab.ust.hk/hs eo/emergncy.htm>

# Plan for Emergencies



---

- What are emergencies which could happen?
- What systems will you need to alert you to an emergency situation?
- What equipment and supplies will you need in the event of an emergency?
- What training will you need?
- Is it safe to work alone in the area?
- What is the phone number for emergency response assistance?



# The Basic Responses for Emergencies (I)

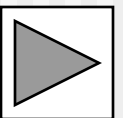
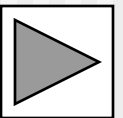
---

- Notify others in the lab 
- Evacuate and isolate the area
- Activate Emergency Ventilation if necessary 
- DO NOT activate EV if there is a fire
- Call Security Control Centre at 8999
- Provide rescue only if you are trained and have proper protective equipment
- Provide first aid assistance

# The Basic Responses for Emergencies (II)

---

- For chemical splash in eyes or on skin:
  - Use eyewash and shower to rinse area for 20 minutes
  - Remove contaminated clothing in the shower
  - Get medical attention
- For small spills, conduct clean up if:
  - Materials is not highly toxic
  - You have supplies
  - You know what you are doing



注 CAUTION 意

HYDROFLUORIC  
ACID

氫 氟 酸

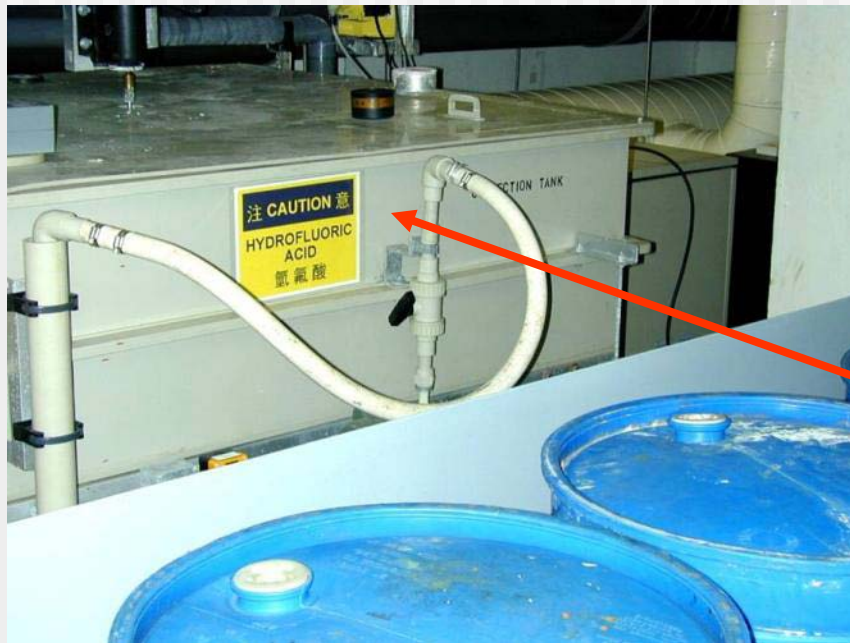
# Hydrofluoric Acid

<b>Concentration</b>	<b>Health Hazard</b>
>50%	<ul style="list-style-type: none"><li>• <i>Immediate burns</i></li><li>• <i>Rapid destruction of tissue/pain</i></li></ul>
20-50%	<ul style="list-style-type: none"><li>• <i>Delayed burns (1-8 hrs)</i></li></ul>
<20%	<ul style="list-style-type: none"><li>• <i>Painful reddening may be delayed up to 24 hrs</i></li></ul>
As low as 2%	<ul style="list-style-type: none"><li>• <i>Delayed skin burns</i></li><li>• <i>Necrosis/slow healing</i></li></ul>





HF waste  
container



Hydrofluoric  
acid storage  
tank

# Hydrofluoric Acid Injury

---

- For low conc exposure, onset of symptoms is often slow and unnoticed
- Damage is often prolonged: for example, contamination around the fingernails may lead to a chronic painful condition and ultimately to amputation of the finger
- Healing is usually slow

# Prepare for Hydrofluoric Acid First Aid

---

- All potential workers and supervisor should be trained in first aid care for HF burns.
- First aid actions should be planned before beginning work with HF.
- Calcium gluconate gel should be readily accessible in areas where HF exposure potential exists.



# Hydrofluoric Acid First Aid

---

1. Immediately shower with large quantities of water (within seconds after contact or suspected contact) - and completely remove all clothing while in the shower (gloves removed last).
2. Flush the skin thoroughly with water for 5 minutes - 5 minutes is sufficient - more time will delay treatment.

# Hydrofluoric Acid First Aid

---

3. Apply calcium gluconate gel (2.5%) at the burn site or area of contamination by rubbing it continuously - wear impervious gloves.
4. Do NOT use calcium gluconate gel for eyes
5. The person should be examined & treated by a physician as soon as possible.

# Information Sources

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# Chemical Information Sources

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- Material Safety Data Sheet (MSDS) associated with each chemical (On-line CCINFODISC and Sigma-Aldrich MSDS Databases)
- Chemical Dictionaries and other references
- Chemical Catalogs
- Labels on reagent bottles





# Labels on Reagent Bottles



Harmful



Harmful and Oxidizing

# Materials Safety Data Sheet

---

- A Material Safety Data Sheet (MSDS) is designed to provide both workers and emergency personnel with the proper procedures for handling or working with a particular substance.

# Materials Safety Data Sheet

---

**MSDS's include the following information:**

- Physical Data
- Risk Phrase,  
Safety Phrase
- Toxicity
- Health Effects
- First Aid
- Reactivity
- Storage
- Disposal
- Protective  
Equipment
- Leak / Spill  
Procedure

# On-Line Chemical Safety Information

---

- International Chemical Safety Cards (by ILO)

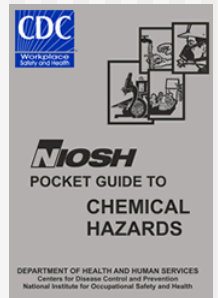
<http://www.ilo.org/public/english/protection/safework/cis/products/icsc/dtasht/index.htm>

- NIOSH Pocket Guide

<http://www.cdc.gov/niosh/npg/npg.html>

- MSDS Search

<http://www.msdssearch.com/DBLinksN.htm>



# Other Chemical Safety Training Courses

---

- *Respiratory Protection*, if you need to use respirators
- *NFF User Safety Training*, if you need to work in MFF
- *Chemical Safety for Operational Staff*

~End of Series~

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Thank you