



Poly(monothiocarbonate)s from the Alternating and Regioselective Copolymerization of COS with Epoxides

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NSFC-RSC
International
Symposium

Hangzhou

27-28, Sept.

Abstract: This poster describes our efforts on the discovery of the selective formation of poly(monothiocarbonate)s from COS with epoxides via heterogeneous zinc-cobalt double metal cyanide complex (Zn-Co(III) DMCC) and homogenous (salen)CrX complexes. [1] The catalytic activity and selectivity of Zn-Co(III) DMCC for COS/epoxide copolymerization are similar to those for CO₂/epoxide copolymerization. COS-based copolymers are highly transparent sulfur-containing polymers with excellent optical properties, such as high refractive index and Abbe number. Very recently, crystalline COS-based polymers with or without chiral carbon have been synthesized, which may further expand the scope of application of these new materials.

Introduction

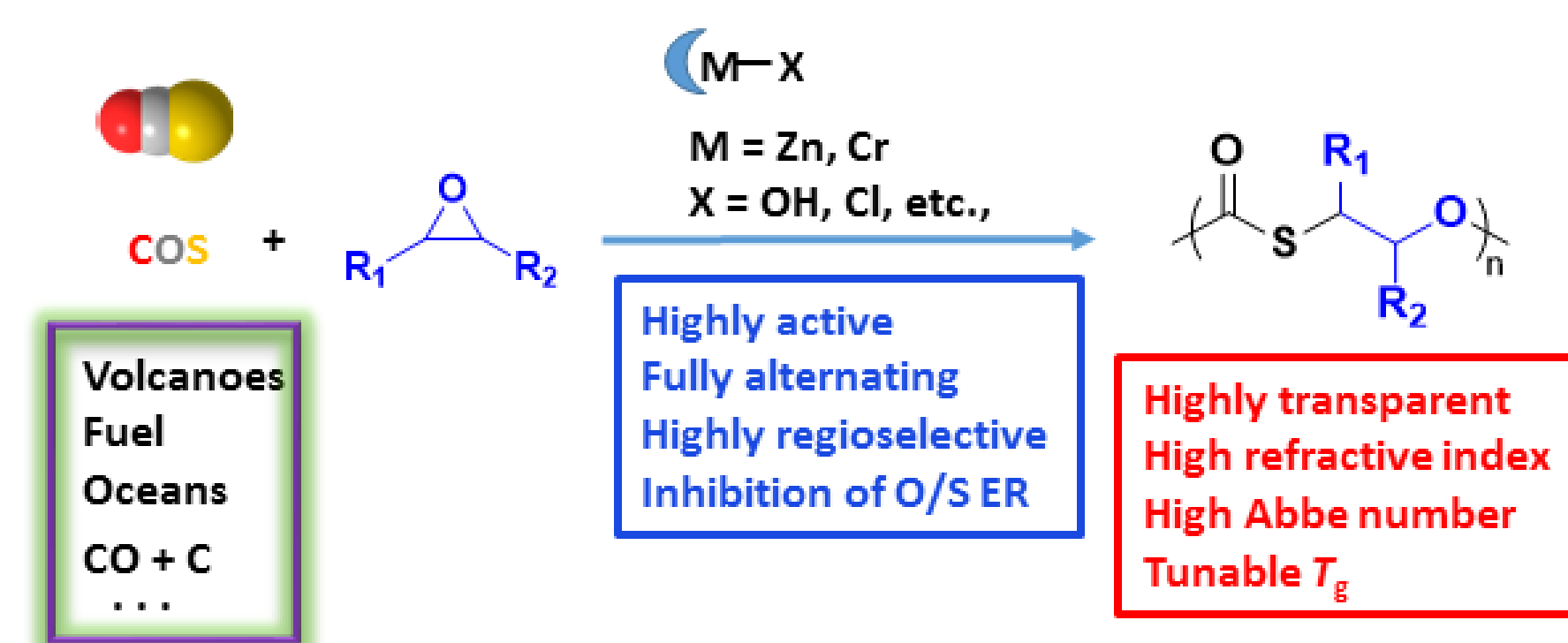


Fig. 1. Synthesis of poly(monothiocarbonate)s from COS/epoxide copolymerization (R₁ and R₂ represent substituted groups).

As an analogue of carbon dioxide (CO₂) and carbon disulfide (CS₂), COS is a heterocumulene containing C=O or C=S groups and can be regarded as an asymmetric form of CO₂ and CS₂. It could be an ideal one-carbon (C1) monomer for synthesizing sulfur-containing polymers via a similar route to CO₂ copolymerization. Motivated by the achievements of CO₂/epoxide copolymerization in the past decade, we have undertaken the research on the COS/epoxide copolymerization for synthesizing poly(monothiocarbonate) (Fig.1).

Catalysts for COS(CS₂)/epoxides copolymerization

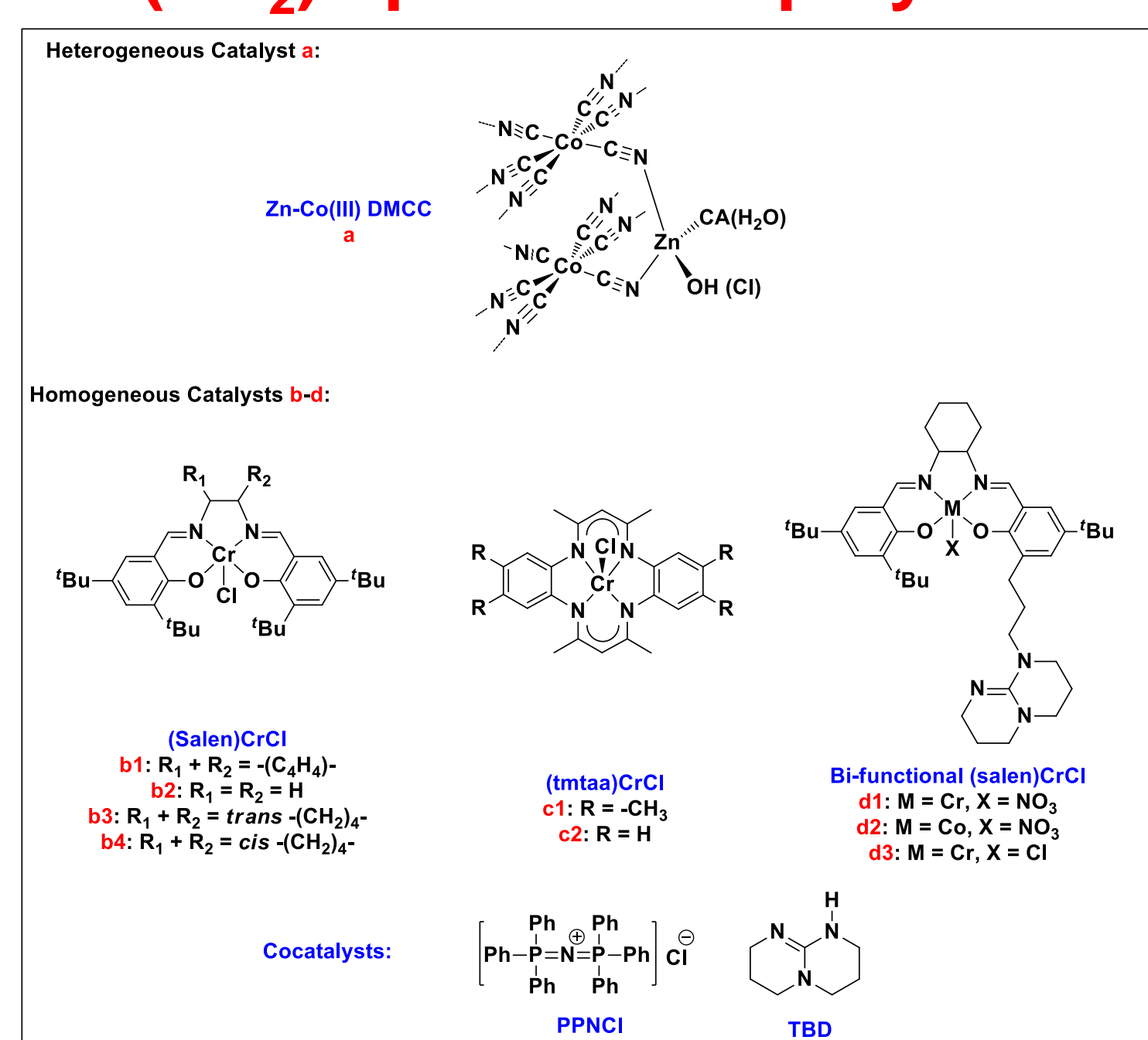


Fig.2 The catalysts and cocatalysts reported for COS/epoxide copolymerization. (The ground state of the catalytic center of a was proposed as a tetrahedral zinc ion with one Zn-OH bond (or Zn-Cl bond), CA is a complexing agent). Cat.d1-d3 were reported by Ren and Lu group. (Macromolecules 2015, 48, 8445-8450)

Since 2008, we have performed the coupling of CS₂ with PO, cyclohexene oxide (CHO), cyclopentene oxide (CPO) and oxetane, by using either heterogeneous zinc-cobalt double cyanide complex [Zn-Co(III) DMCC] catalyst or homogenous (salen)CrCl complex (Fig.2). **Oxygen-sulfur exchange reaction (O/S ER)** was observed in both the polymeric and cyclic products resulting in irregular polymer chain structure and poor selectivity.

The observation of COS intermediate during CS₂/epoxides copolymerization intrigued us to explore COS/epoxides copolymerization. The catalytic activity and selectivity of Zn-Co(III) DMCC for COS/epoxide copolymerization are similar to those for CO₂/epoxide copolymerization. (Salen)CrX complexes accompanied by onium salts exhibited high activity and selectivity for COS/epoxide copolymerization under mild conditions, affording copolymers with >99% monothiocarbonate units and a high tail-to-head content up to 99%. The specialty of COS polymerization that are the unit selectivity caused by O/S ER and regioselectivity induced by the asymmetrical structure of COS. Of importance, COS can copolymerize with numerous epoxides, whether they contain electron-donating or electron-withdrawing groups, displaying high catalytic activities and regioselectivity in a well-controlled process carried out under mild conditions.

Conclusions

The production of sulfur-containing polymers from the copolymerization of COS and epoxides represents an atom-economical and significantly “greener” route to these thermoplastics. Our efforts on the COS/epoxides copolymerization disclosed the specialty of COS polymerization that are the unit selectivity caused by O/S ER and regioselectivity induced by the asymmetrical structure of COS. With regard to future directions for synthesizing COS-based copolymers, we hope to broaden the scope of catalysts capable of catalyzing COS with epoxides, affording various poly(monothiocarbonate)s, as well as their block copolymers and crystalline polymers.

Acknowledgement

X.-H.Z. gratefully acknowledge the financial support of the Distinguished Young Investigator Fund of Zhejiang Province (LR16B040001) and the National Science Foundation of the People's Republic of China (no. 21274123). D.J.D. is thankful to the Robert A. Welch Foundation (A-0923) for the financial support.

Mechanistic aspects of the COS/epoxide copolymerization

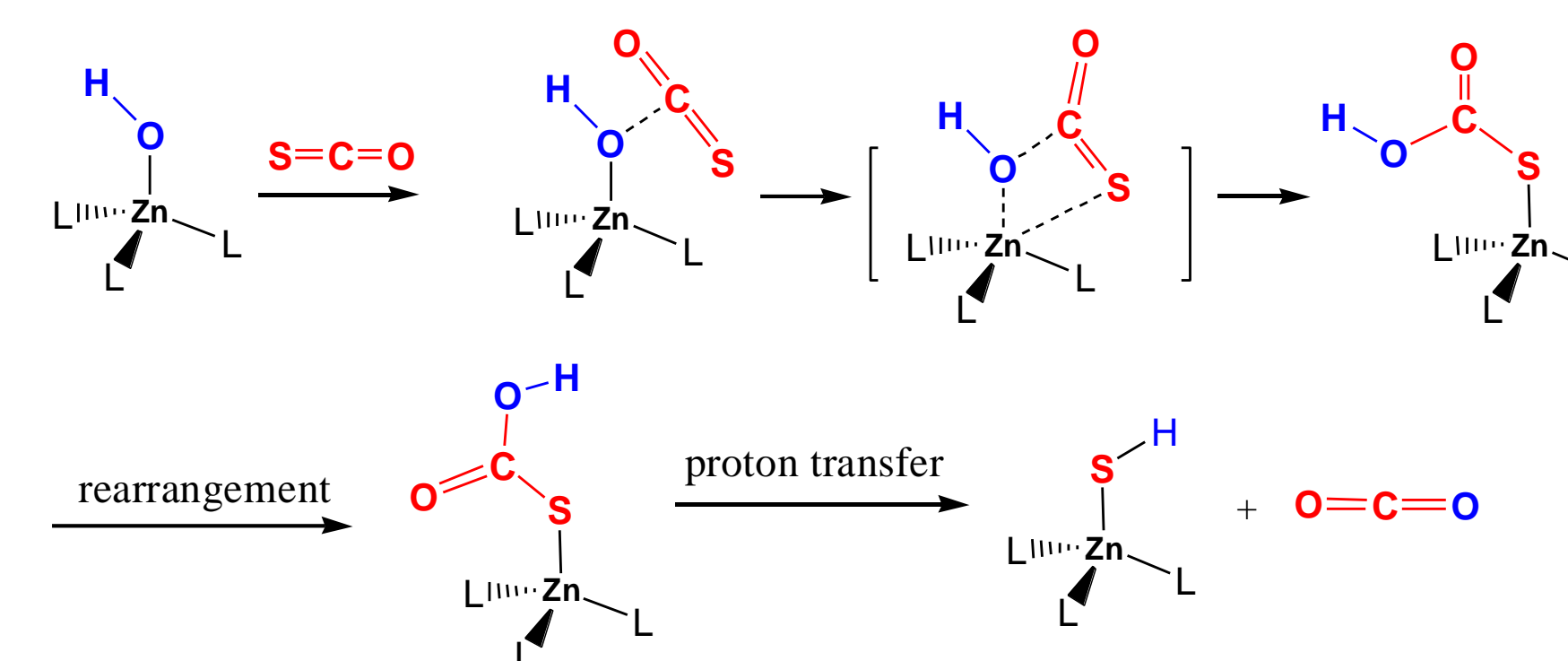


Fig. 3. Metal-OH catalysis for O/S ER. Similar to the mechanism of fixation of COS by the CA compound [L₃ZnOH]⁺, L is histidine (CA center structure) or ligand (generally, amine, CA models).

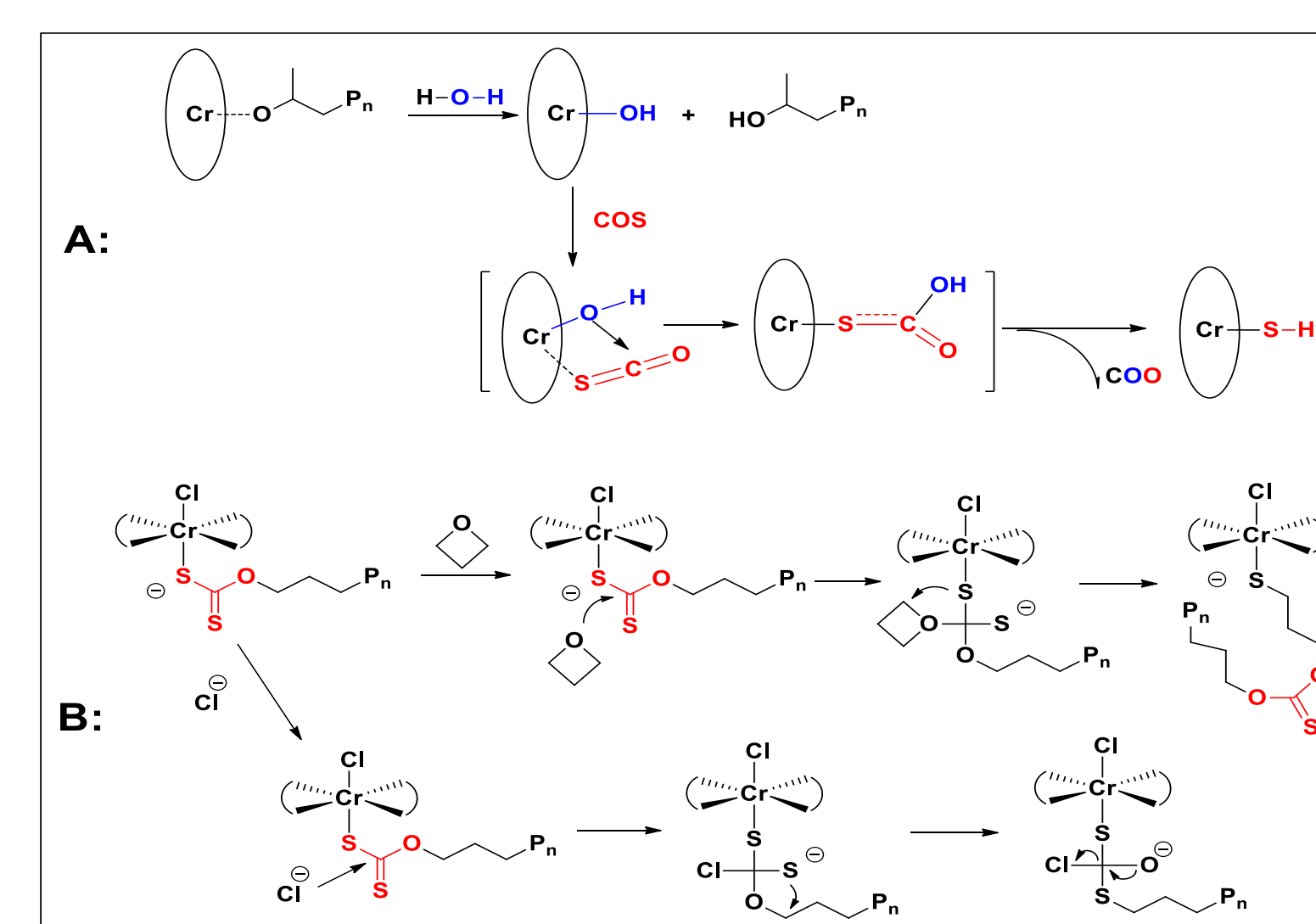


Fig. 4. Proposed O/S ER mechanism for COS/epoxide and CS₂/oxetane copolymerization. (A) Metal-OH catalysis, (B) Rearrangement of tetrahedral intermediate (Take Cr³⁺ as an example).

Properties of COS/epoxide copolymer

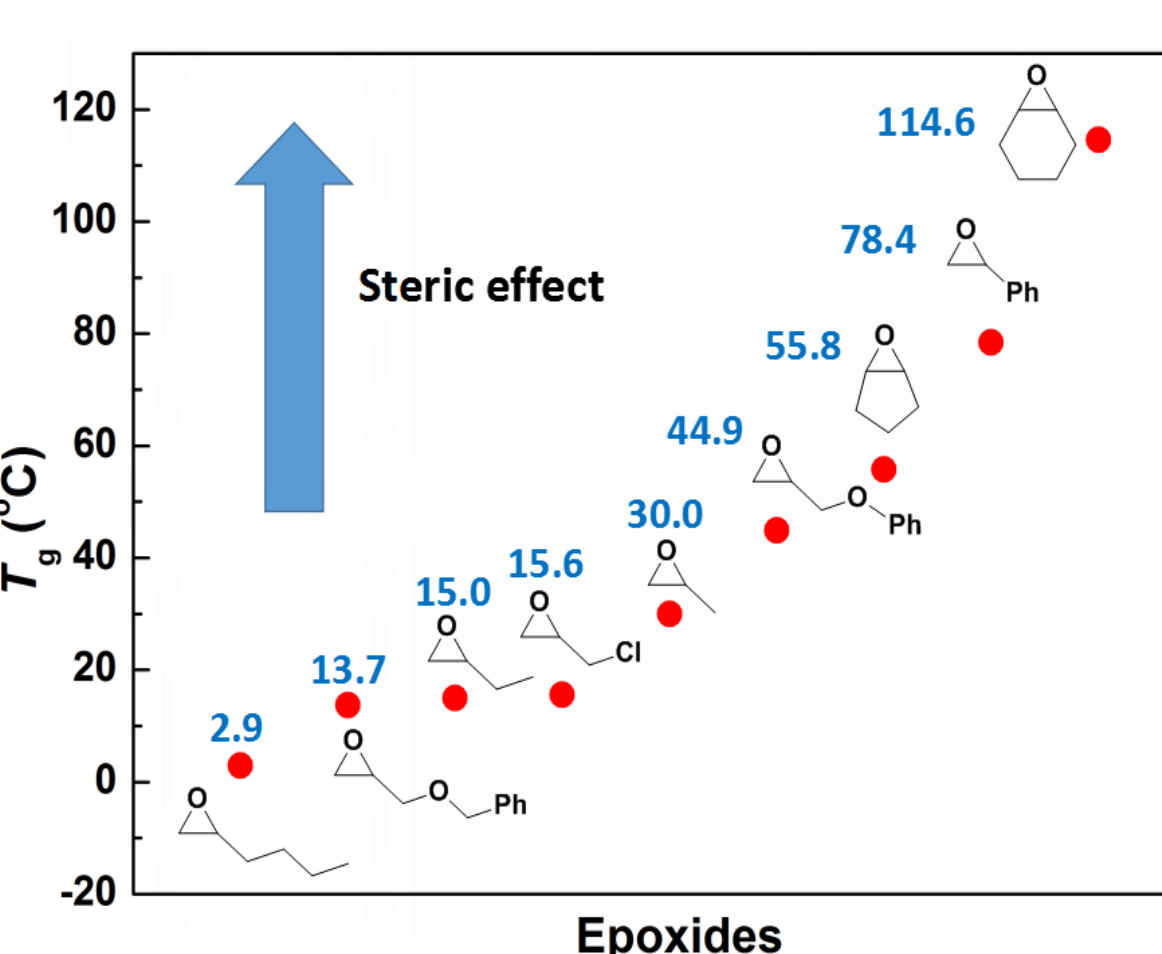


Fig. 5. T_gs of the poly(monothiocarbonate)s from COS/epoxides copolymerization.

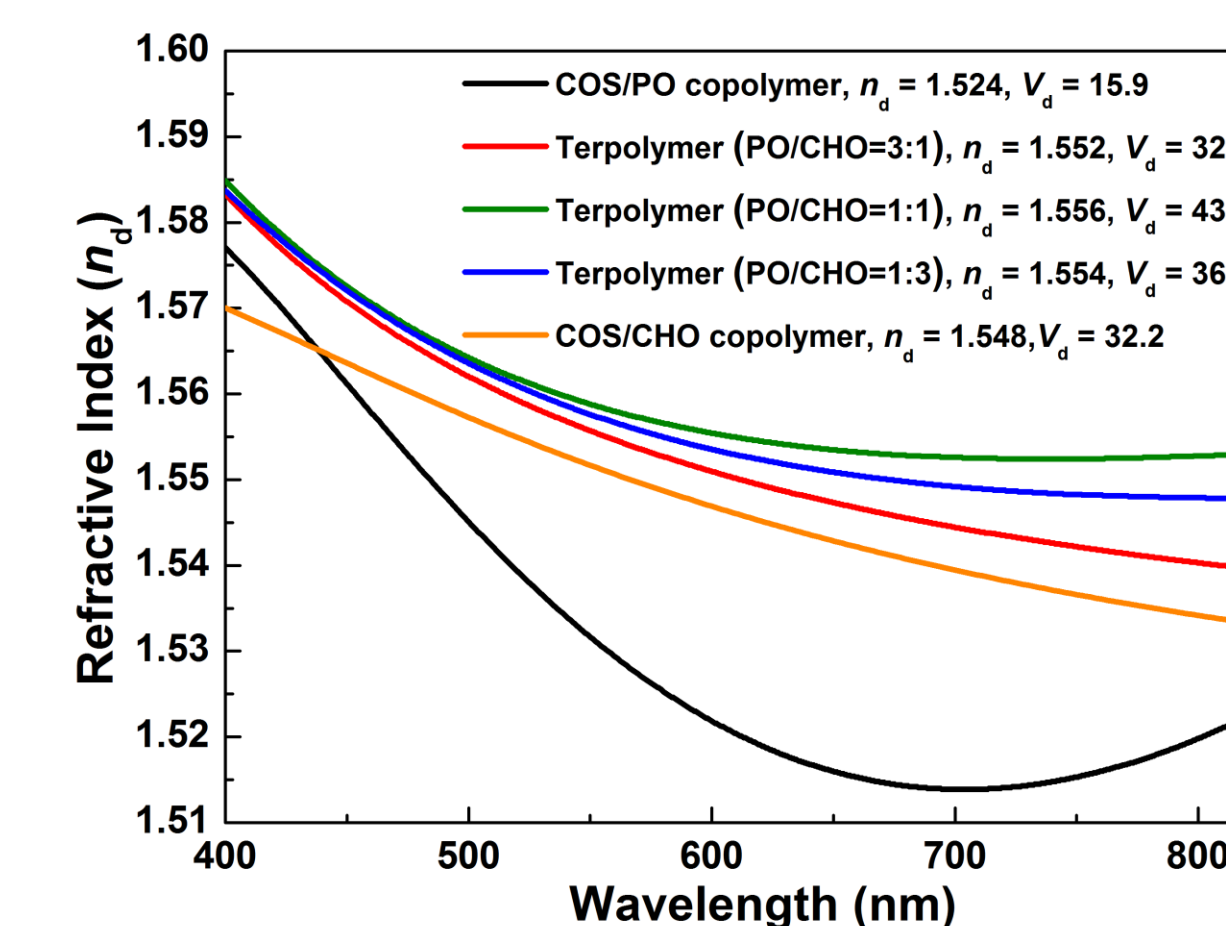


Fig. 6. The refractive indices (n) of COS/CHO/PO terpolymers via the wavelength from 400 to 800 nm (n_D is refractive index at wavelength 587.6 nm).

Crystalline COS/oxetane copolymer

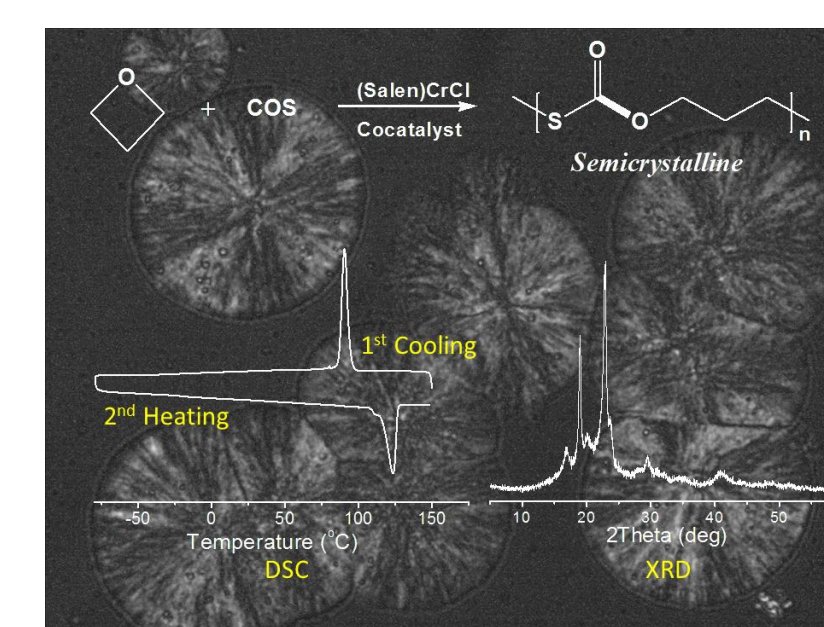


Fig.7 A semicrystalline poly(trimethylene monothiocarbonate) (PTMTC) was firstly synthesized via the selective and alternating copolymerization of COS with oxetane. It presents similar crystallization behavior to high-density polyethylene (HDPE), i.e., being spherulite and possesses melting temperature (T_m) of up to 127.5°C and degree of crystallinity (X_c) of up to 71%. [2]

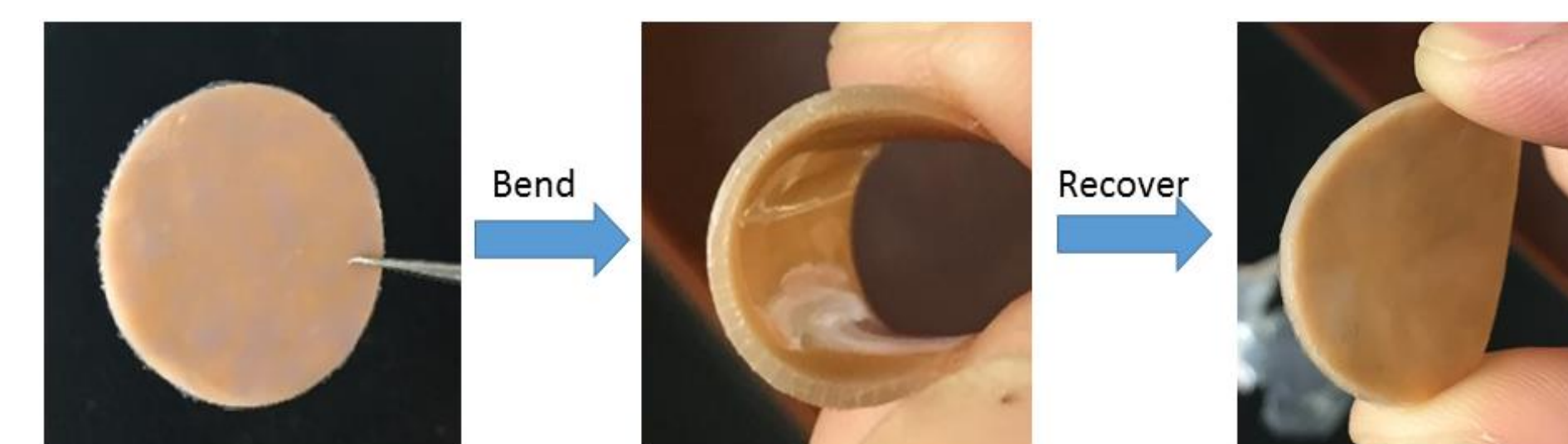


Fig.8 PTMTC could be hot pressed at 150°C for 8 min and then cool pressed at room temperature for 8 min. It is flexible and tough.

References

- [1] Luo, M.; Zhang, X. H.; Darensbourg, D. J. Poly(monothiocarbonate)s from the Alternating and Regioselective Copolymerization of COS with Epoxides. (Accepted)
- [2] Wu H. L.; Yang J. L.; Luo, M.; Wang R. Y.; Xu J. T.; Du B. Y.; Zhang, X. H.; Darensbourg, D. J. Poly(trimethylene monothiocarbonate) from Alternating Copolymerization of COS and Oxetane: A Semicrystalline Copolymer. (Submitted)