# Catalyst-Free Thiol-Yne Click Polymerization: A Powerful <br> and Facile Methodology for Preparation of Functional Poly(vinylene sulfide)s 



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## INTRODUCTION

The "thioclick" polymerization is a well expanded concept of click polymerization. ${ }^{1}$ Among these polymerizations, the thiol-yne click polymerization is less developed and still in its infancy stage. In general, UV light, elevated temperature, amine or transition metal complexes are needed to catalyze the thiol-yne polymerization, which greatly complicates the experimental operation and limits its application. ${ }^{2}$ Here, we developed a facile and powerful catalyst-free thiol-yne click polymerization for the preparation of functional poly(vinylene sulfide)s (PVSs). ${ }^{3}$

## Thiol-Yne Click Polymerization

Scheme 1. Syntheses of poly(vinylene sulfide)s by catalyst-free thiol-yne click polymerizations of diynes $\mathbf{1}$ and dithiols $\mathbf{2}$ or $\mathbf{3}$.


Table 1. The click polymerizations of diynes 1 with dithiols 2-4. ${ }^{\text {a }}$

| No. monomers polymer |  |  |  | yield <br> $(\%)$ | $M_{w}{ }^{\mathrm{b}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |$M_{\mathrm{w}} / M_{\mathrm{n}}{ }^{\mathrm{b}}$.

${ }^{\text {a }}$ Carried out in THF at $30{ }^{\circ} \mathrm{C}$ under nitrogen for 2 h ; $[\mathrm{M}]_{0}=50 \mathrm{mM} .{ }^{b}$ Estimated by GPC in THF. ${ }^{c} 4=1,5-$ pentanedithiol.

## RESULTS AND DISCUSSION



Figure 1. ${ }^{1} \mathrm{H}$ NMR spectra of monomers (A) 1a and (B) 2, (C) model compound, and (D) polymer Pla in $\mathrm{CDCl}_{3}$.

## Mechanism Study

Table 2. The polymerization of $\mathbf{1 a}$ with $\mathbf{2}$ in the presence of $\gamma$-terpinene. ${ }^{\text {a }}$

| No. | [add] <br> $(\mathrm{mM})^{\mathrm{b}}$ | yield <br> $(\%)$ | $\mathrm{S}^{\mathrm{c}}$ | $M_{\mathrm{w}}{ }^{\mathrm{d}}$ | $M_{\mathrm{w}} / M_{\mathrm{n}}{ }^{\mathrm{d}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 0 | 97 | $\sqrt{ }$ | 60900 | 1.98 |
| 2 | 50 | 63 | $\sqrt{ }$ | 1400 | 1.50 |
| 3 | 100 | 53 | $\sqrt{ }$ | 1000 | 1.36 |

${ }^{\text {a }}$ Carried out in THF at $30{ }^{\circ} \mathrm{C}$ under nitrogen for 2 h ; $[\mathrm{M}]_{0}=50 \mathrm{mM} .{ }^{b}$ add $=\gamma$-terpinene. ${ }^{c}$ Solubility (S) tested in common organic solvents; $\sqrt{ }=$ completely soluble. ${ }^{d}$ Estimated by GPC in THF.

Light Refractivity


Figure 2. Light refraction spectra of thin solid films of polymers Pla-Ple and PII.

## Aggregation-Induced Emission



Figure 3. (A) PL spectra of Ple in THF and THF/water mixtures. Polymer concentration: 10 $\mu \mathrm{M}$. Excitation wavelength: 330 nm . (B) Plot of relative PL intensity versus water fraction in THF/water mixtures, where $I=$ peak intensity and $I_{0}=$ peak intensity in pure THF. Inset: fluorescent images taken under a hand-held UV lamp.

## ACKNOWLEDGMENTS

## CONCLUSIONS

 catalyst-free click polymerization of aromatic diynes and dithiols under mild conditions. The PVSs possess good solubility and thermal stability, and excellent light refractivity in film state. The tetraphenylethene containing polymer PIe exhibits the unique aggregation-induced emission feature..This work was supported by the National Science Foundation of China (21222402 and 21174120) and the Ministry of Science and Technology of China (2013CB834702).

## REFERENCES

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