

**Fabrication of novel polyelectrolyte complex membranes containing strong-acid groups for pervaporation dehydration** 

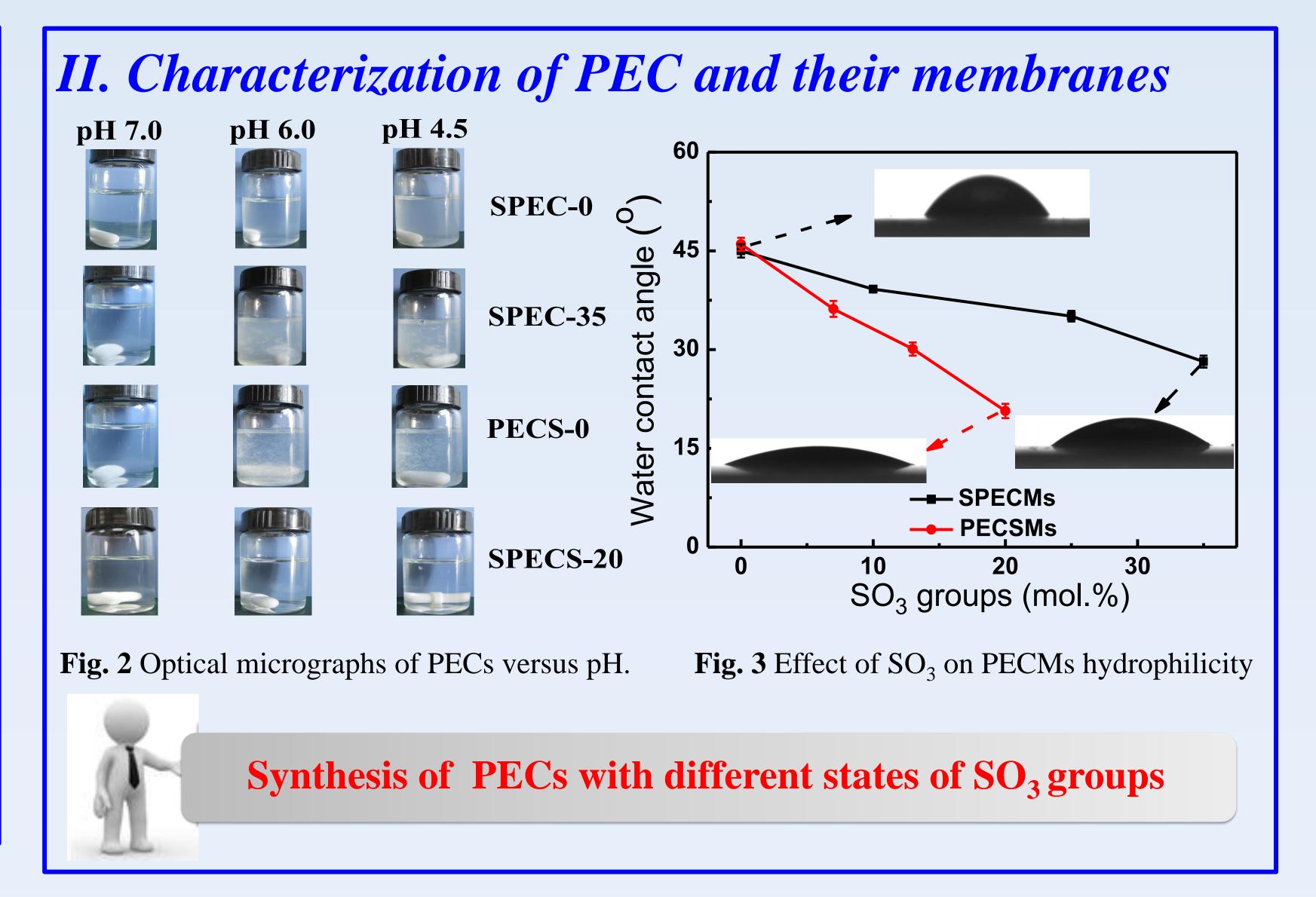


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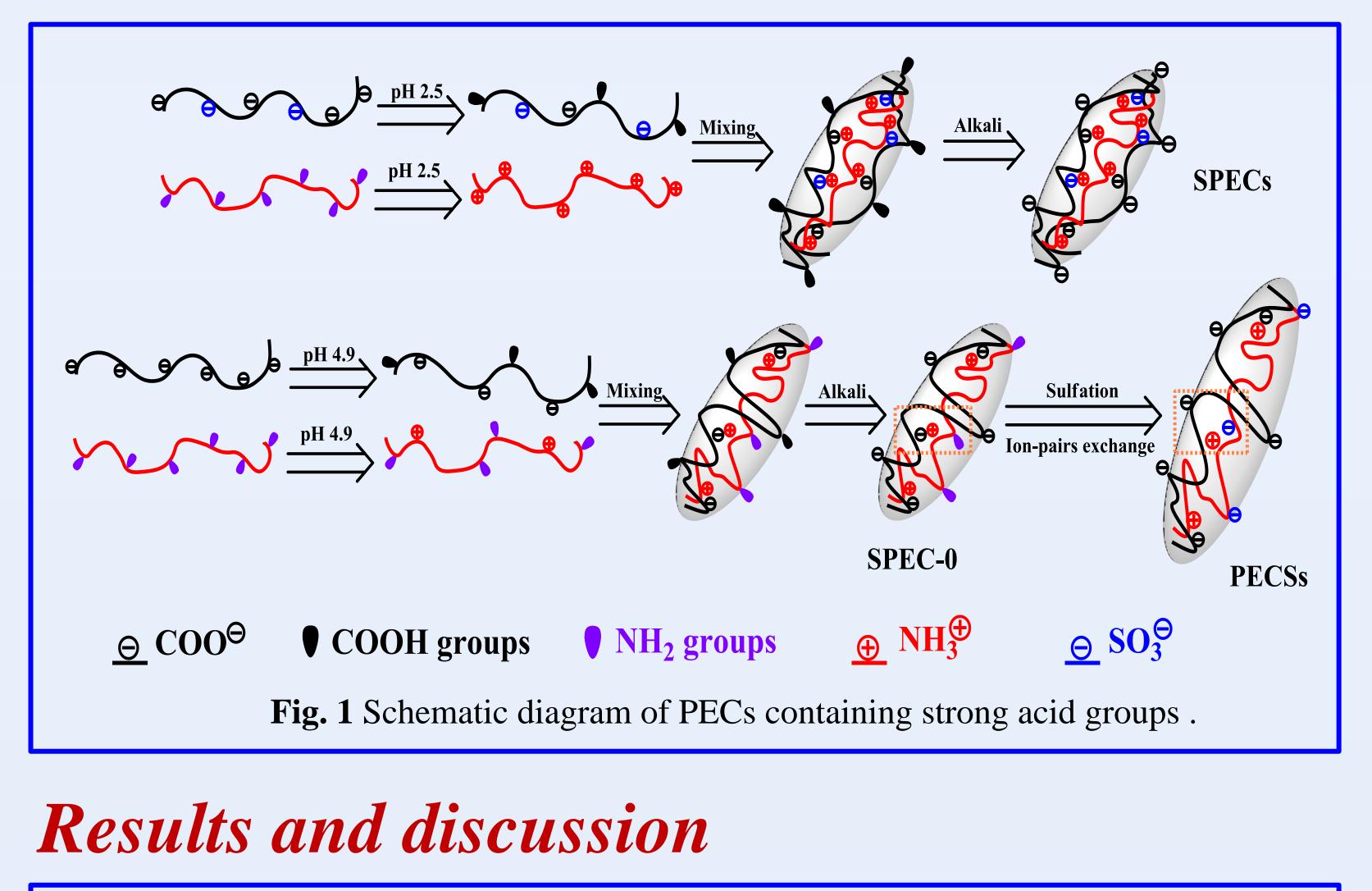
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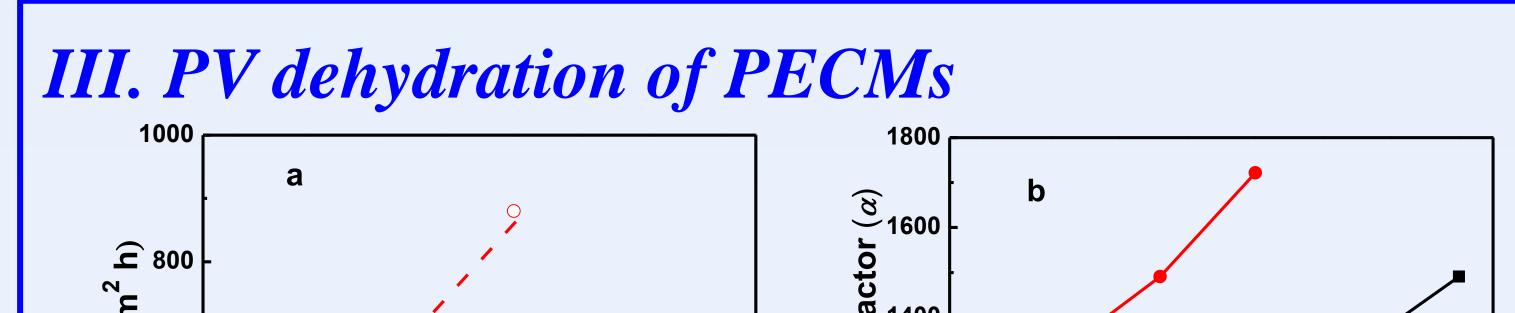
## Introduction

Polyelectrolyte complexes (PECs) are blended at the molecular level and classified as a type of multi-component polymeric materials. PECs have been performed as a promising candidate for membrane (PECM) application ranging from pervaporation (PV), nanofiltration to the proton exchange membrane as well as gas separation. These homogenous PECMs made from soluble PECs exhibited very high PV performance in organics dehydration. Owing to the high hydrophilicity and strong interaction formed by sulfate (SO<sub>3</sub>) groups [1], the SO<sub>3</sub> groups could enhance the water channels embedded in PECMs hydrophilicity and mechanical properties [2]. In this work, the PECMs containing different states of SO<sub>3</sub> groups were fabricated. It was found that the free SO<sub>3</sub> groups was capable of effectively improving the PV performance while the complexed ones enhancing the mechanical properties of PECMs.



# Experimental





I. Characterization of PECs

Table 1 Composition of PECs determined by XPS						
Sample	S (At. %)	N (At. %)	S:N	Х	Corresponding PECM	
SPEC-0		1.57	0	0	SPECM-0	- <b>PECMs with</b>

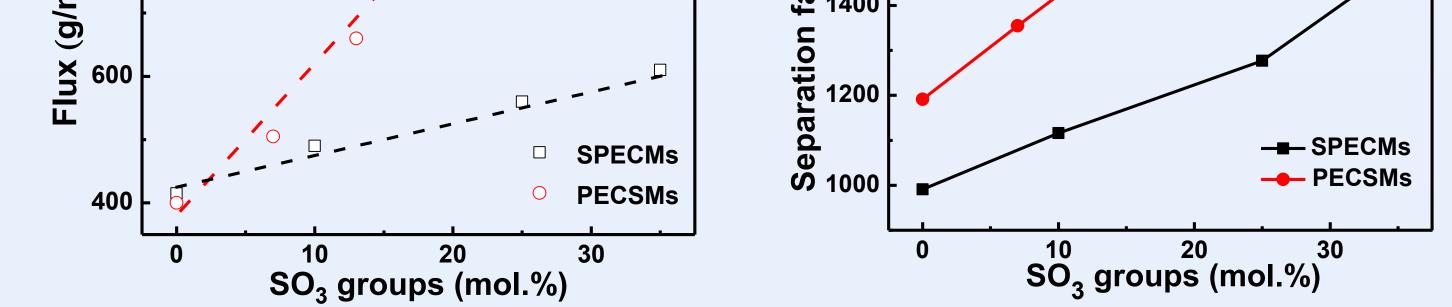


Fig. 4 Effect of SO<sub>3</sub> content on the flux (a) and separation factor ( $\alpha$ ) (b) for PECMs in the dehydration of 10 wt. % water-ethanol mixtures at 50 °C.

### **IV. Mechanical properties of PECMs**

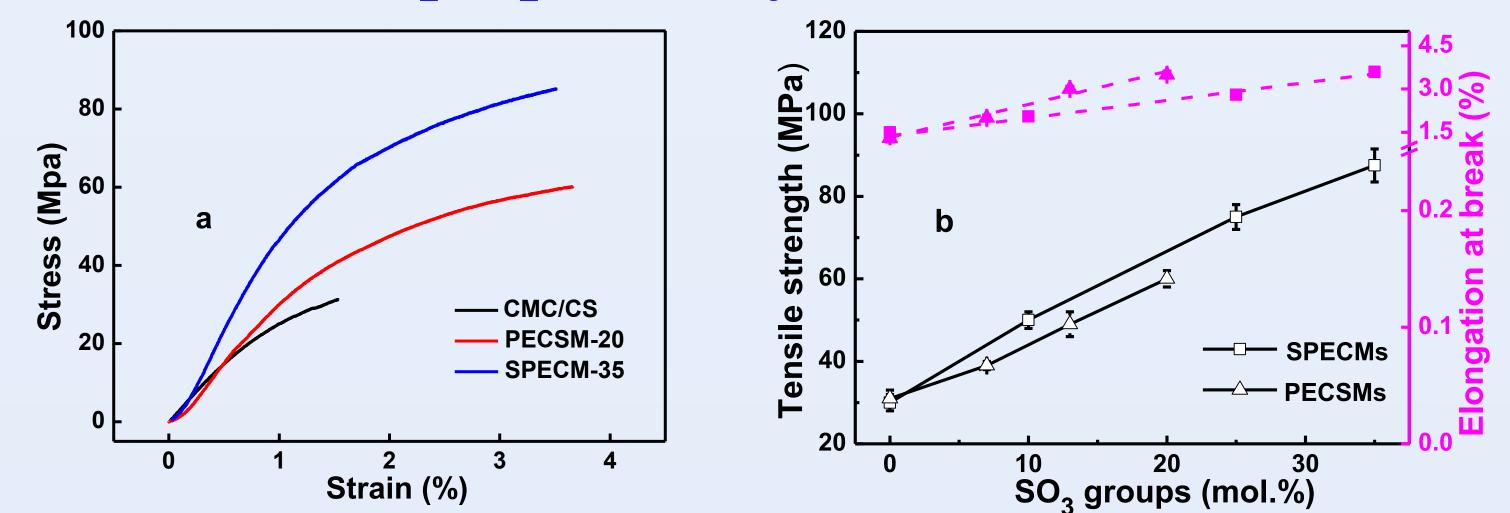
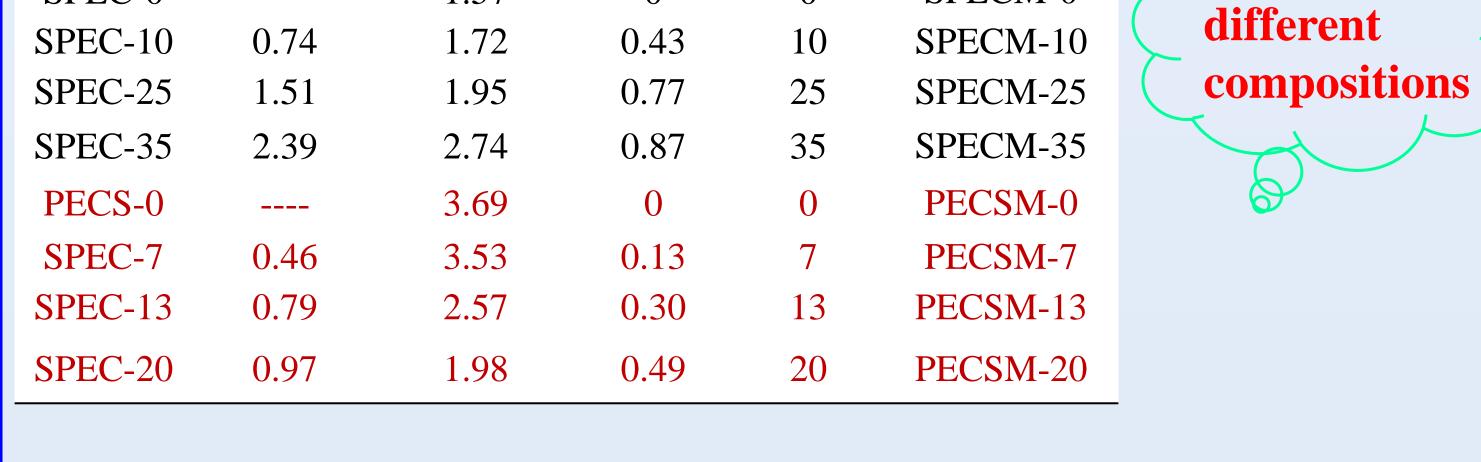


Fig. 5 Typical stress-strain curves of PECMs based on different states of  $SO_3$  groups (a) and effect of  $SO_3$  content on of the mechanical properties of PECMs (b).

### Conclusions



Synthesis of novel PECMs containing SO<sub>3</sub> groups.
Free SO<sub>3</sub> effectively improved PECMs PV dehydration.
Complexed SO<sub>3</sub> availably enhanced PECMs mechanical properties.

## References

[1] D.V. Pergushov, A.H.E. Müller, F.H. Schacher, *Chem. Soc. Rev.* 2012, *41*, 6888. [2] X.S. Wang, Q.F. An, T. Liu, etc., *J. Membr. Sci.* 2014, *452*, 73.

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